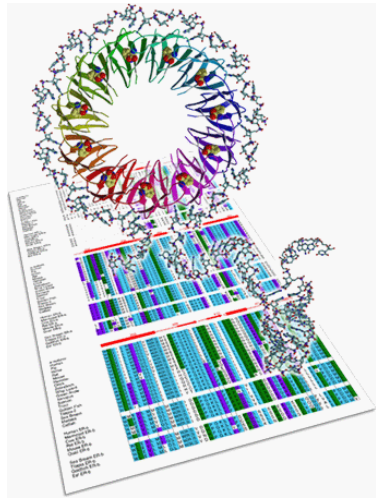


Escuela Complutense de Verano Especialista en Bioinformática



Interacciones entre proteínas y pequeños ligandos (I)

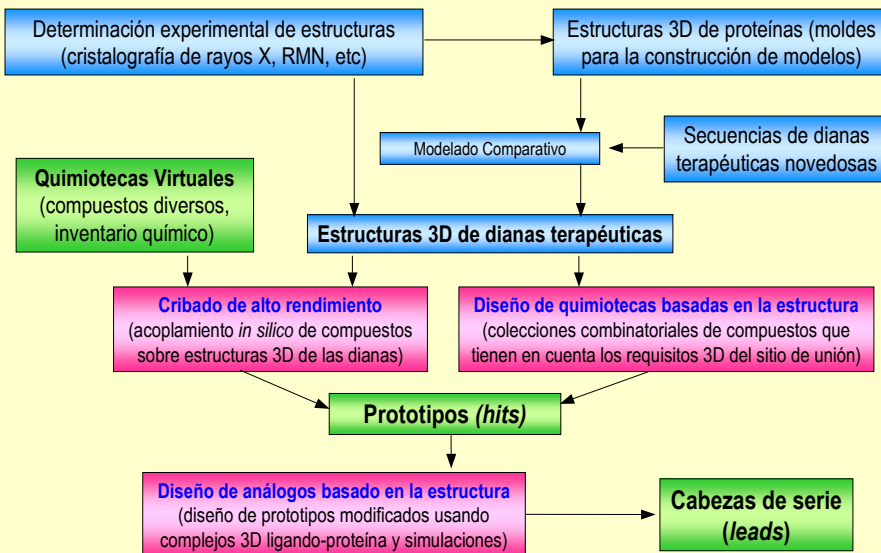
Federico Gago
Departamento de Farmacología
Universidad de Alcalá, Madrid



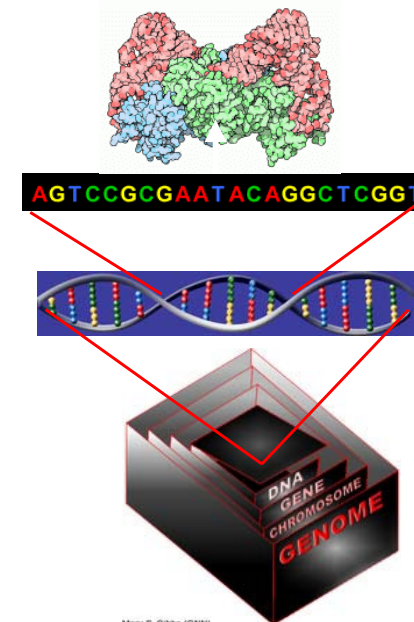
Interacciones entre proteínas y moléculas pequeñas

1. Concepto de ligando y sitio de unión. Ejemplos.
 2. Bases de datos estructurales y programas asociados.
 3. Caracterización estructural de moléculas pequeñas y sus complejos con proteínas.
 4. Acoplamiento ligando-receptor ("docking"): algoritmos y programas.
 5. Cribado virtual.
-
6. Relaciones estructura-actividad: QSAR y 3D-QSAR.
 7. Diseño de nuevos ligandos.

Integración de Metodologías Informáticas y Basadas en la Estructura para el Descubrimiento de Fármacos



Gene expression = Protein production



How Proteins Work

Proteins recognize and reversibly bind to other molecules: *cofactors, substrates, inhibitors...* Also *ions* and other *proteins*.

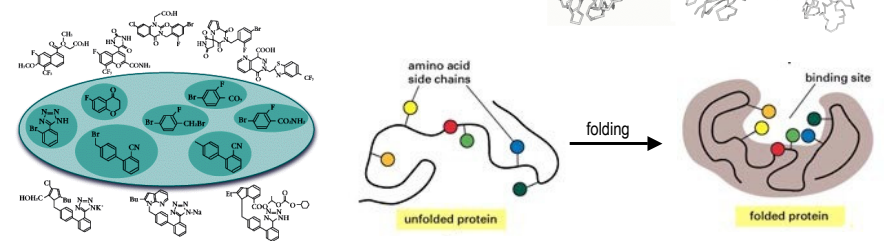
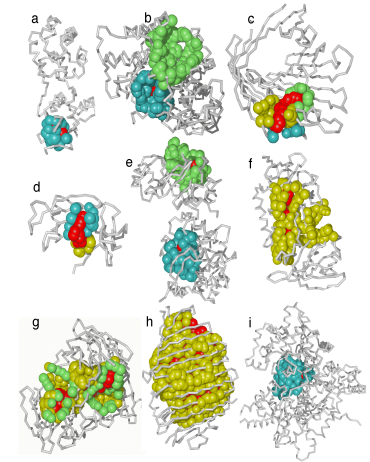
The bound small molecule is called a **ligand**.

In the case of **enzymes**, the region that associates with substrates and products is called the **active site**.

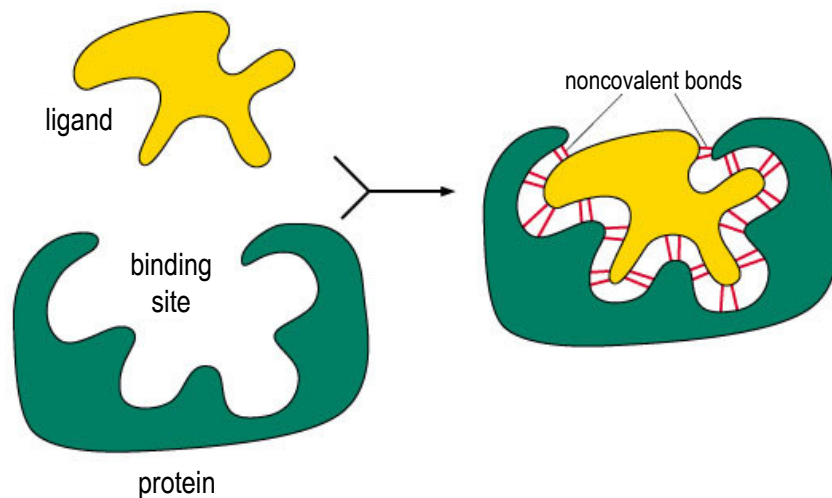
The region of a protein that associates with activator or inhibitor molecules is called an **allosteric site**.

Proteins can have > 1 binding site for different ligands

Proteins fold in such a way that they create **specific sites** that are the right *size*, *shape*, and *polarity* for their ligands.

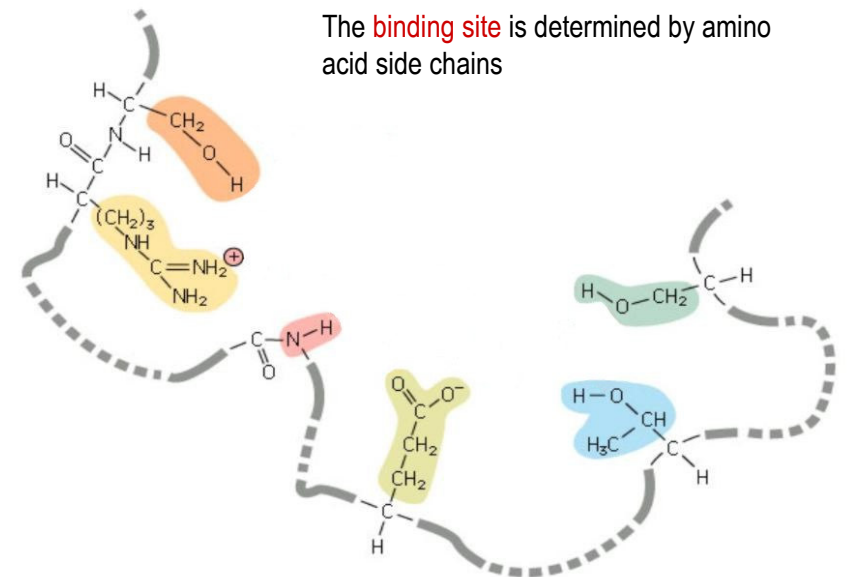


Ligand binding is highly selective

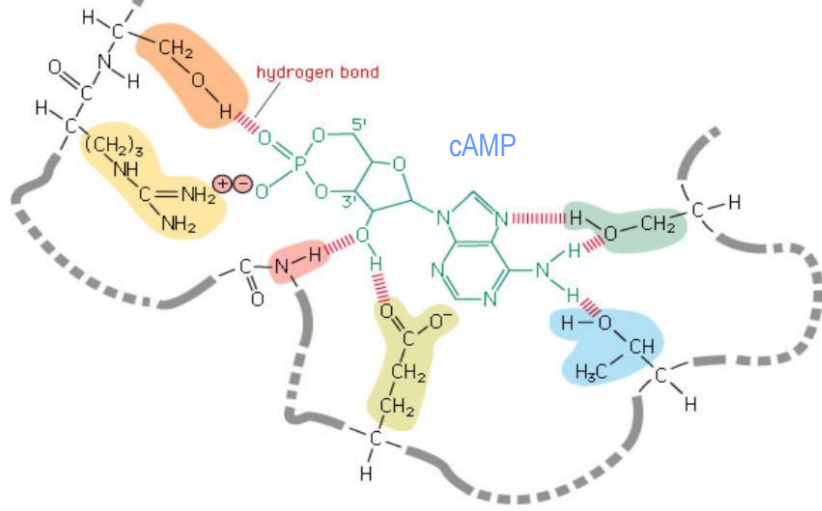


©1998 GARLAND PUBLISHING

The **binding site** is determined by amino acid side chains



Non covalent interactions stabilize the complex formed between the ligand and the binding site

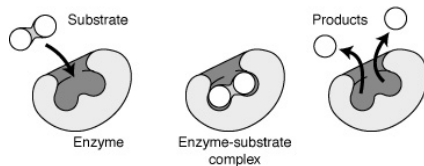


Sitios de unión: complementariedad de forma



... y complementariedad electrostática

Enzymes: A special case of protein function

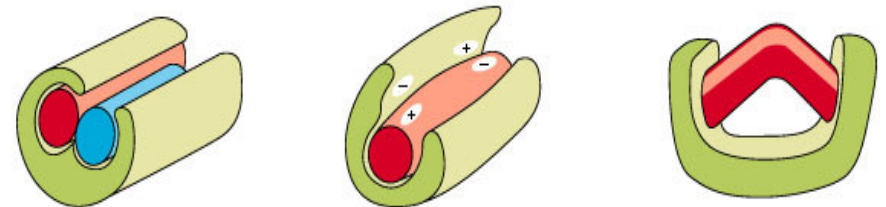


Enzymes bind and assist in the chemical transformation of other molecules

Substrate: molecule acted upon by an enzyme (analogous to ligand)

Catalytic site: substrate-binding site (analogous to ligand-binding site)

How do enzymes catalyze reactions?

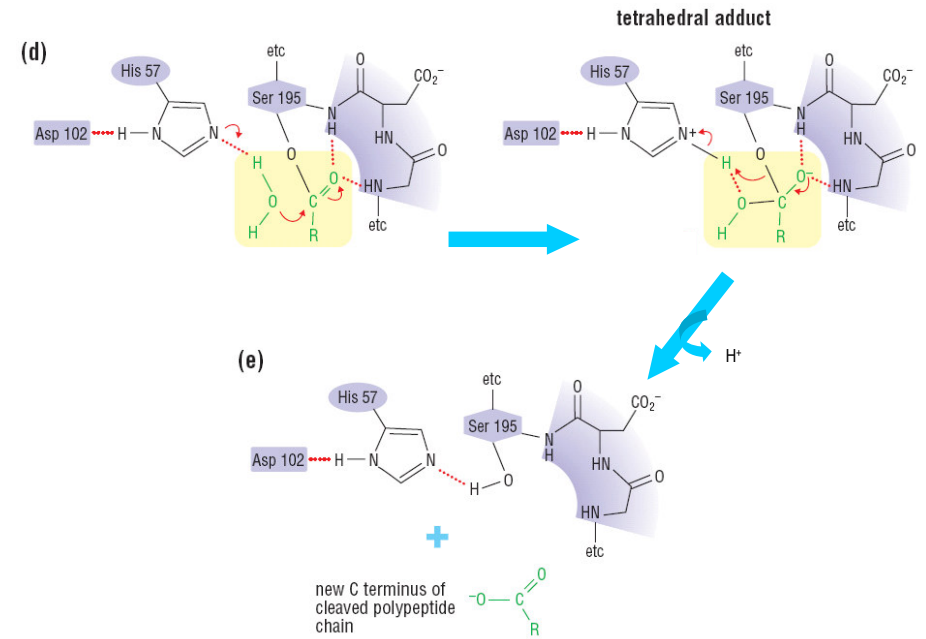
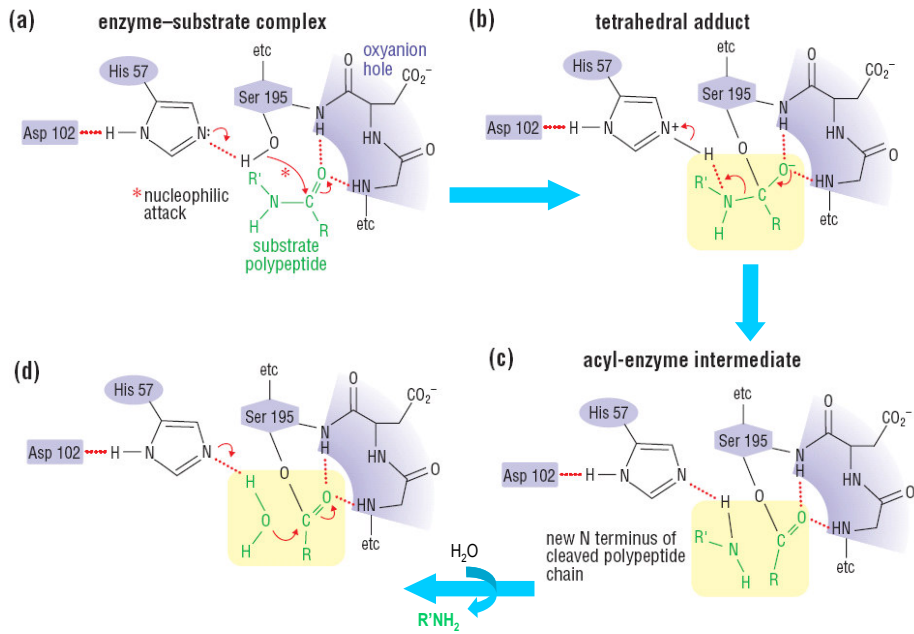


(A) enzyme binds to two substrate molecules and orients them precisely to encourage a reaction to occur between them

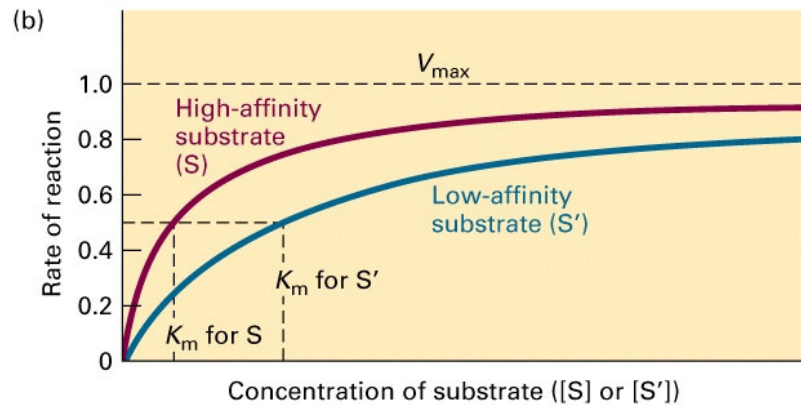
(B) binding of substrate to enzyme rearranges electrons in the substrate, creating partial negative and positive charges that favor a reaction

(C) enzyme strains the bound substrate molecule, forcing it toward a transition state to favor a reaction

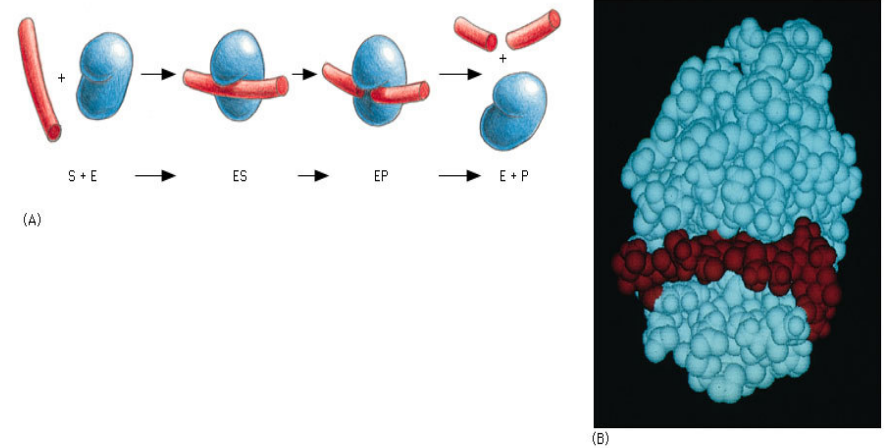
The chemical steps in peptide hydrolysis catalyzed by the **serine protease chymotrypsin**



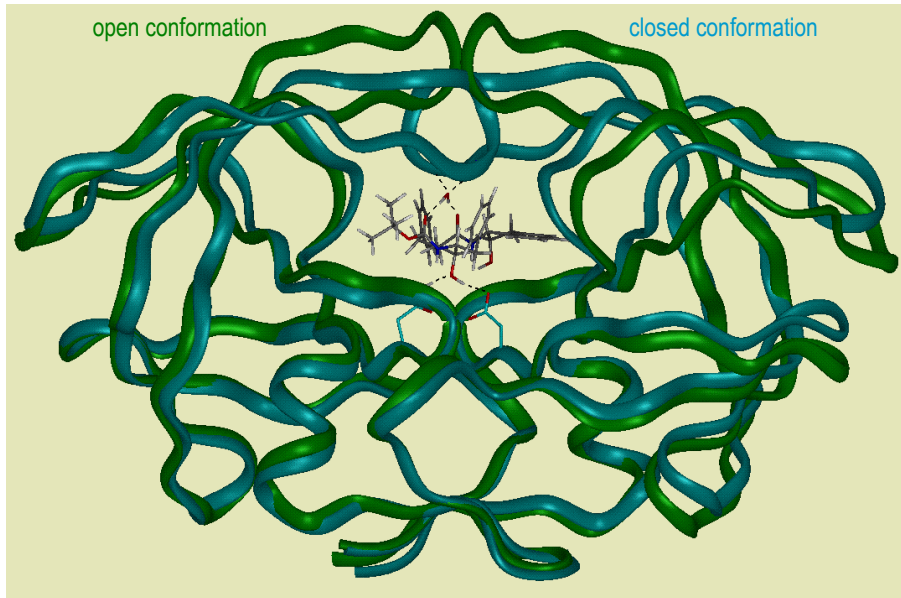
Kinetics of an enzymatic reaction are described by V_{max} and K_m



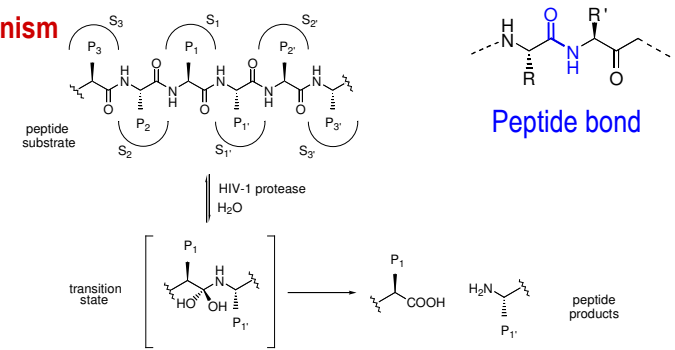
Lysozyme catalyzes the cutting of a **polysaccharide** chain



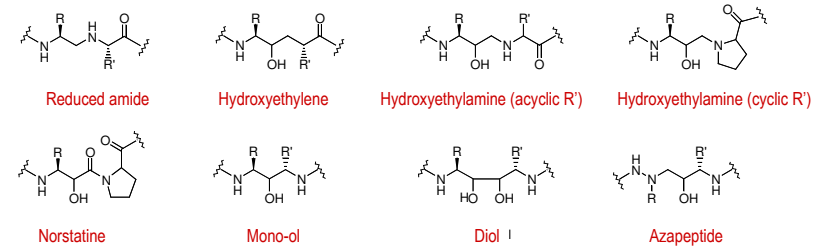
HIV-1 protease



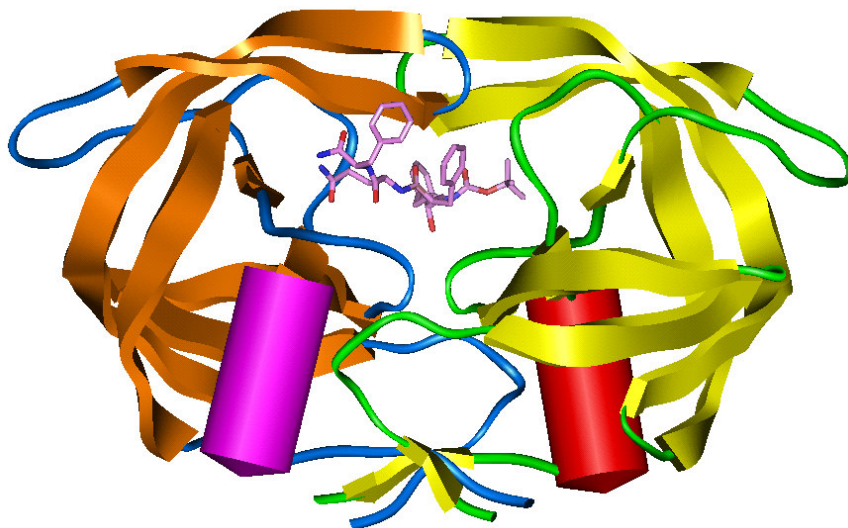
Hydrolytic mechanism in HIV-1 protease



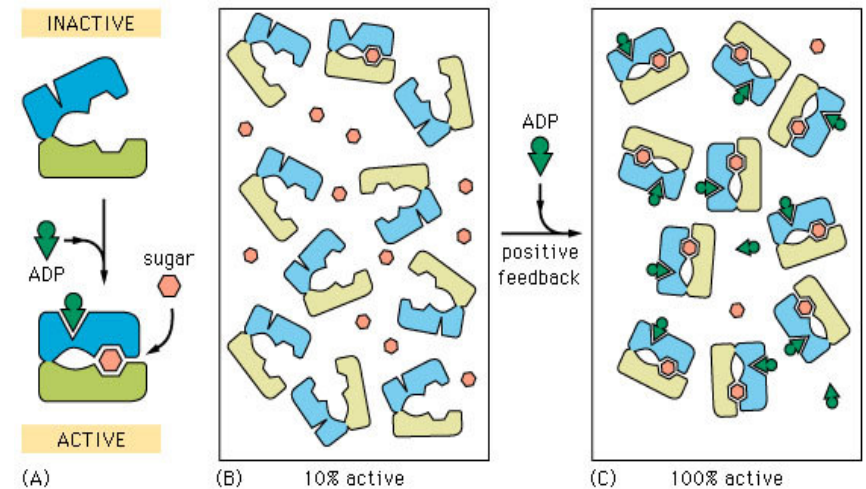
Examples of non-hydrolyzable **isosteres** of the **peptide bond** cleaved by HIV-1 protease



HIV-1 protease in complex with inhibitor QF-34

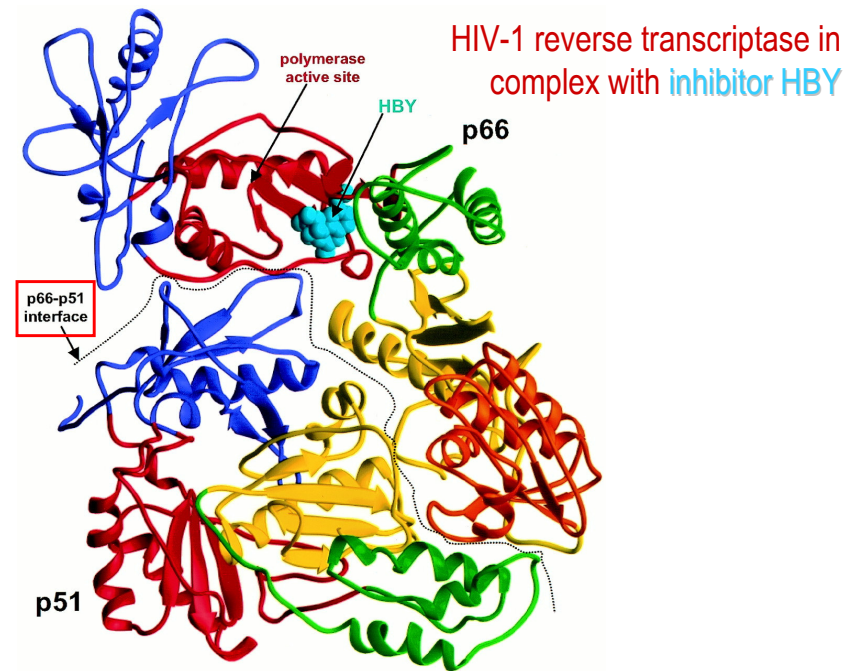
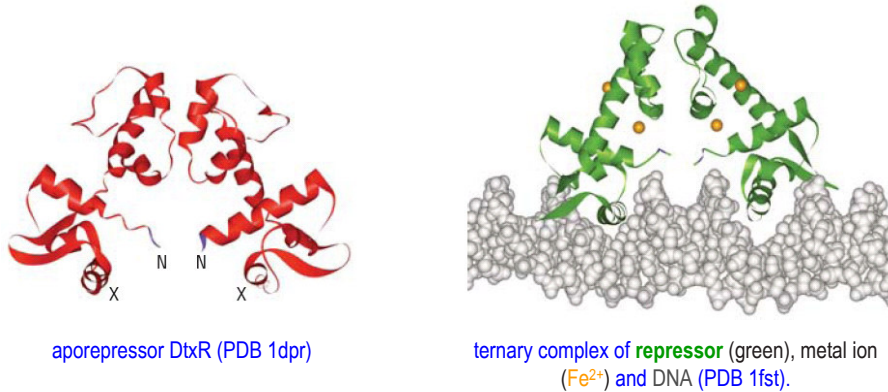


Enzyme activation caused by an allosteric change



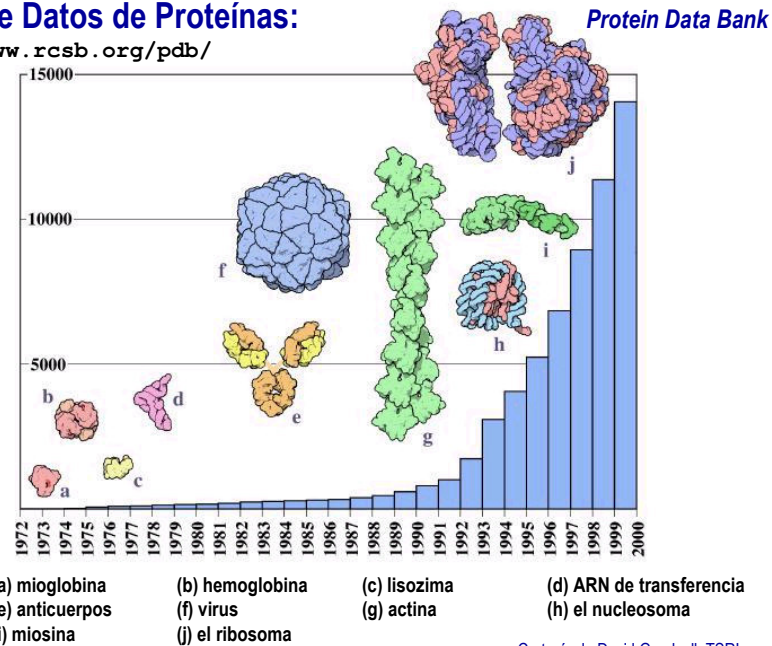
Binding of gene regulatory proteins to DNA is often controlled by ligand-induced conformational changes

Iron binding regulates the repressor of the diphtheria toxin gene:



Banco de Datos de Proteínas:

<http://www.rcsb.org/pdb/>

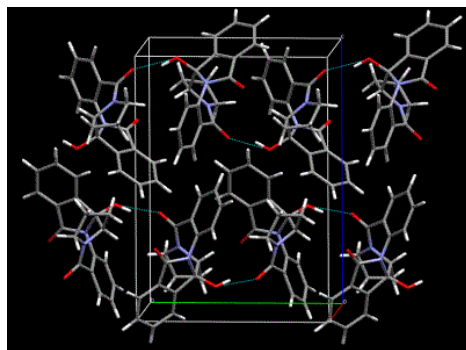


Cambridge Structural Database

The Cambridge Crystallographic Data Centre (CCDC) builds, maintains and distributes the Cambridge Structural Database (CSD), a searchable database of organic and metallo-organic crystal structures.

The CCDC also produce and distribute software products which make use of the data contained in the CSD.

Increasing the Value of Crystallographic Databases



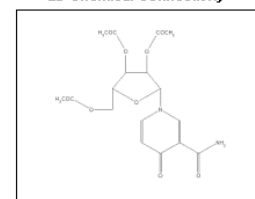
- Derived knowledge bases
- Knowledge-based applications programs
- Data mining tools for protein-ligand complexes



1D Bibliographic Information

BASY0J
 4-Oxonicotinamide-1-
 (1'-beta-D-2',3',5'-tri-O-
 acetyl-ribofuranoside)
 Source: *Rochmannia longiflora*
 C17 H20 N2 O9
 G. Bringmann, M. Ochse, K. Wolf,
 J. Kraus, K. Peters, E-M. Peters,
 M. Herderich, L. Ake, F. Tayman
Phytochemistry 51 (1999), p271

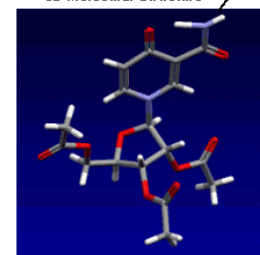
2D Chemical Connectivity



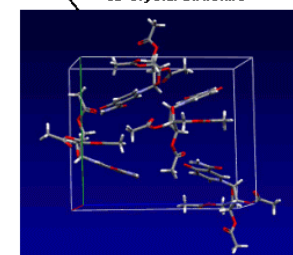
>272,000 organic and metallo-organic
 crystal structures analysed using
 X-ray or neutron diffraction techniques

CSD

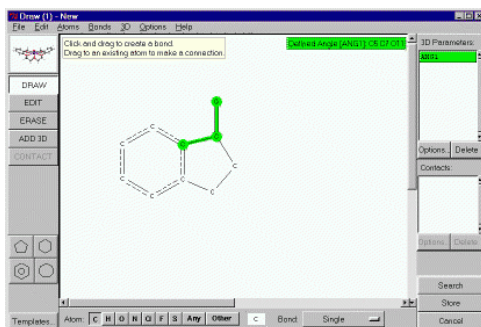
3D Molecular Structure



3D Crystal Structure

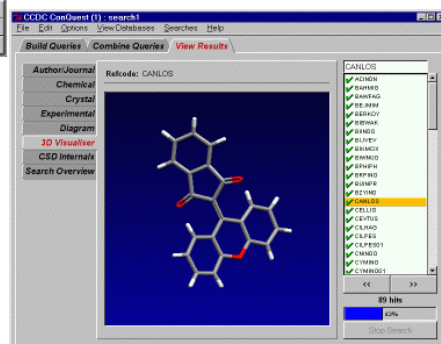


ConQuest



ConQuest provides a full range of text/numeric database search options, in addition to more complex search functionality, including:

- Chemical substructure searching
- Geometrical searching
- Intermolecular non-bonded contact searching



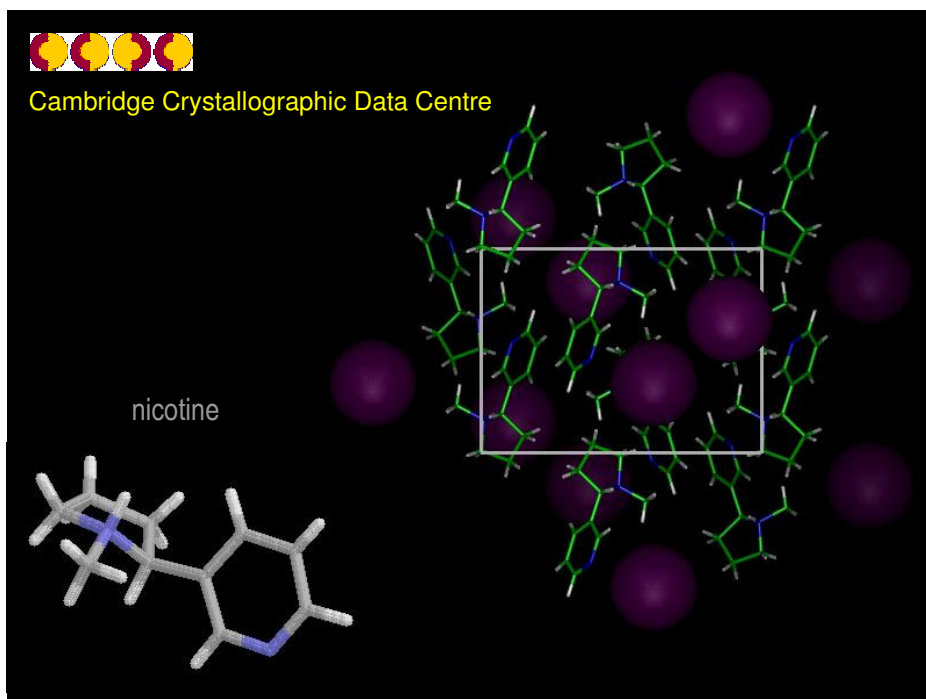
Cambridge Crystallographic Data Centre

<http://www.ccdc.cam.ac.uk/>

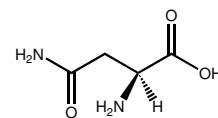
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#DOXSIS 33870428      16 9 0 0 0 4 4 28 0 0 301322000001000000000086
1370
R=0
211
CRYST 13.705      9.974      8.436  90.00  90.00  P212121
ATOM 1 I1 NICO 1      1.583  1.561  0.355  1.00  0.00
I1
C2
ATOM 2 N1 NICO 1      5.355 -5.377 10.252  1.00  0.00
C2
C5
ATOM 3 C1 NICO 1      5.744 -4.111 10.057  1.00  0.00
C5
C8
ATOM 4 C2 NICO 1      5.375 -3.343  8.953  1.00  0.00
C8
C10
ATOM 5 C3 NICO 1      4.573 -3.927  7.982  1.00  0.00
C10
H3
ATOM 6 C4 NICO 1      4.180 -5.249  8.151  1.00  0.00
H3
H6
ATOM 7 C5 NICO 1      4.584 -5.922  9.302  1.00  0.00
H6
H9
ATOM 8 C6 NICO 1      5.840 -1.905  8.899  1.00  0.00
H9
H12
ATOM 9 C7 NICO 1      4.752 -0.841  8.748  1.00  0.00
H12
H15
ATOM 10 C8 NICO 1      5.461  0.374  8.127  1.00  0.00
H15
ATOM 11 C9 NICO 1      6.826 -0.125  7.655  1.00  0.00
0
ATOM 12 N2 NICO 1      6.722 -1.627  7.699  1.00  0.00
....
CONECT 1 0
CONECT 2 3 7
CONECT 3 2 4 14
....
MASTER 0 0 0 0 0 0 0 0 28 0 28 0
END
```



Cambridge Crystallographic Data Centre

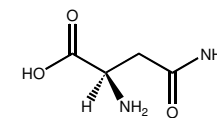


The Importance of Chirality

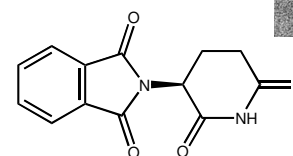
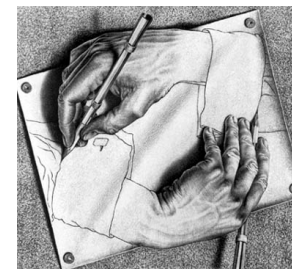


Bitter

Asparagine

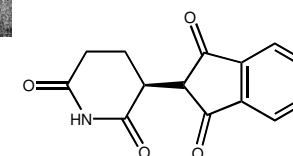


Sweet



Extreme teratogen

Thalidomide



Anti-morning sickness

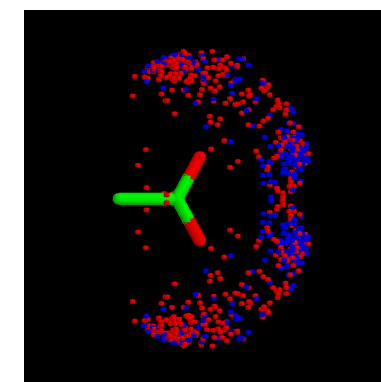
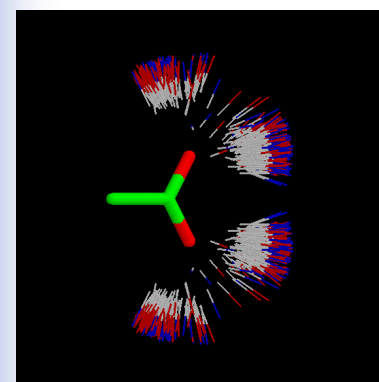


IsoStar and **SUPERSTAR**

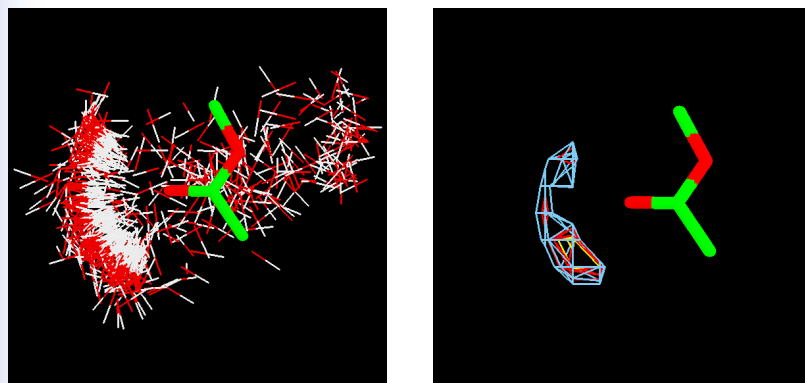
- **IsoStar** - knowledge base of information about intermolecular interactions
- **SuperStar** - program for predicting binding points in an enzyme active site
- **SuperStar** predictions based solely on **IsoStar** data



IsoStar Scatterplots



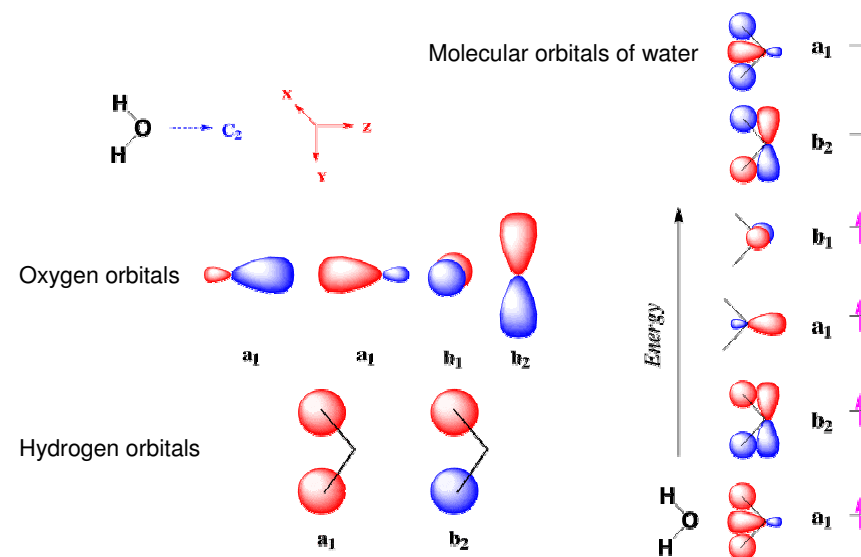
IsoStar Density Surfaces



A probability surface derived from the observed positions of hydrogen-bonding hydrogen atoms around aliphatic esters.

Quantum Chemistry

Atomic orbitals can be combined to give molecular orbitals



Ab initio METHODS

* [Hartree-Fock method](#)

* [Electron correlation methods](#)

▪ **variational methods**

Configuration Interaction with double excitations (CID)

Configuration Interaction with single and double excitations (CISD)

▪ **perturbation methods**

Møller and Plesset (MP2, MP3, MP4)

Quadratic Convergence CI method (QCISD)

▪ **density functional methods (DFT)**

BP86 - developed by Becke and Perdew in 1986

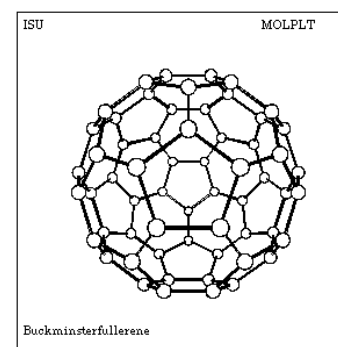
BLYP - developed by Becke, Lee, Yang and Parr

B3LYP - a modification of BLYP in which a 3-parameter functional developed by Axel Becke is used.

GAMESS

General Atomic and Molecular
Electronic Structure System

<http://www.msg.ameslab.gov/GAMESS/GAMESS.html>

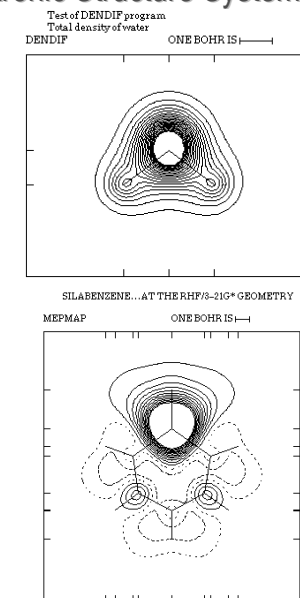


```

Enter red only!
Color numbers:
1234:87:9:103:45
Control Command
Draw again
Write restart
Quit

rotations
  Y
  X
  Z
  clockwise sense
X <angle> <n>
Y <angle> <n>
Z <angle> <n>

Reset Options
ATOM <symbol>
<kolor> <size>
BW <bandwidth>
BL <badlength>
V <viewdistance>
S <scalemode>
    
```





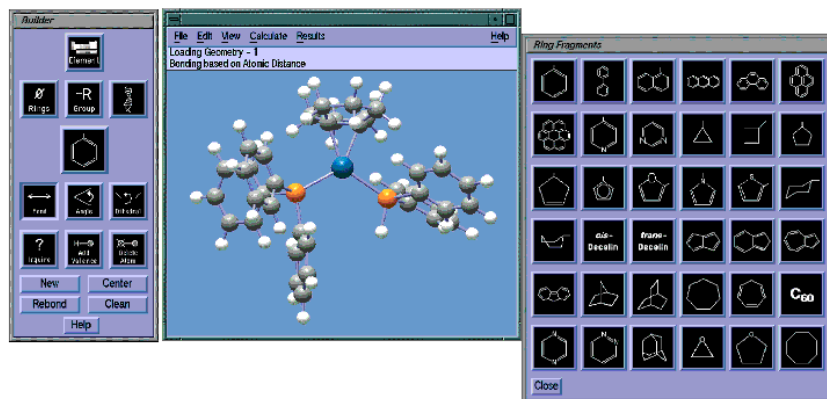
Gaussian

<http://www.gaussian.com/>



Spartan

<http://www.wavefun.com/>



Some sample Gaussian z-matrices

Water (C_{2v})



With variables:

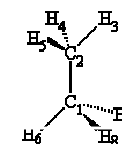
```
o
h 1 11
h 1 11 2 a1
```

```
11 0.96
a1 104.0
```

With values:

```
o
h 1 0.96
h 1 0.96 2 104.0
```

Ethane (D_{3d})



```
c
c 1 11
h 2 12 1 a1
h 2 12 1 a1 3 120.0
h 2 12 1 a1 3 -120.0
h 1 12 2 a1 3 180.0
h 1 12 2 a1 6 120.0
h 1 12 2 a1 6 -120.0
```

```
11 1.54
12 1.09
a1 110.0
```

SEMI-EMPIRICAL METHODS: levels of approximation

CNDO Complete Neglect of Differential Overlap (Developed by John Pople - assumes atomic orbitals to be spherical when evaluating the two-electron integrals)

INDO Intermediate Neglect of Differential Overlap

NDDO Neglect of Diatomic Differential Overlap

MINDO/3 Modified INDO (Developed by Michael Dewar - uses a set of parameters to approximate the two-electron repulsion integrals)

ZINDO Includes parameters for transition metals

MNDO Modified NDO (Developed by Michael Dewar and Walter Thiel in 1977)

AM1 Austin Model 1 (Developed by Michael Dewar and Andrew Holder in 1986)

PM3 Parametric Model 3 (Developed by Jimmy Stewart in 1988)

Sample input for MOPAC

```
PM3 EF PRECISE
```

```
H2O (water)
```

```
MOPAC input as a Z-matrix
```

```
o
H          0.96000  1
H          0.96000  1    104.00000  1    1  2
```

Water (C_{2v})



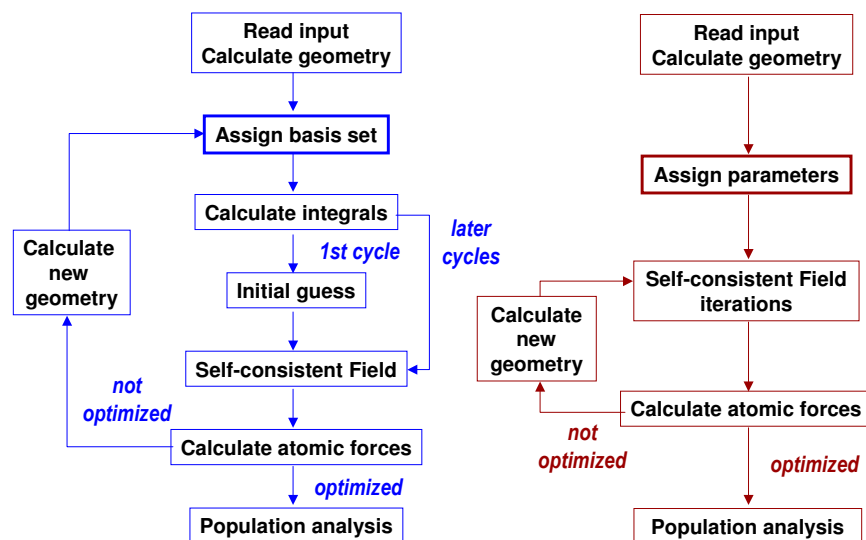
```
AM1 EF PRECISE
```

```
H2O (water)
```

```
MOPAC input in Cartesian coordinates
```

```
o          0.0000  0  0.0000  0  0.0000  0
H          0.9600  1  0.0000  0  0.0000  0
H          -0.2322  1  0.9315  1  0.0000  0
```

Typical flow charts for an *ab initio* optimization and a corresponding semi-empirical calculation



SMILES

Simplified Molecular Input Line Entry Specification

Rules

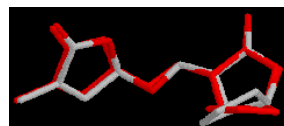
1. Atoms are represented by atomic symbols: B, C, N, O, F, P, S, Cl, Br, and I.
2. Double bonds are '=', triple bonds are '#'.
3. Branching is indicated by parentheses.
4. Ring closures are indicated by pairs of matching digits.

Examples

Depiction	SSMILES	Name	Remark
	C	methane	hydrogens fill normal valence
	CCO	ethanol	a single bond is assumed to join adjacent atoms
	CC(=O)O	acetic acid	parentheses are used to indicate branching
	C1CCCCC1	cyclohexane	bonds can also be represented by pairs of matching digits

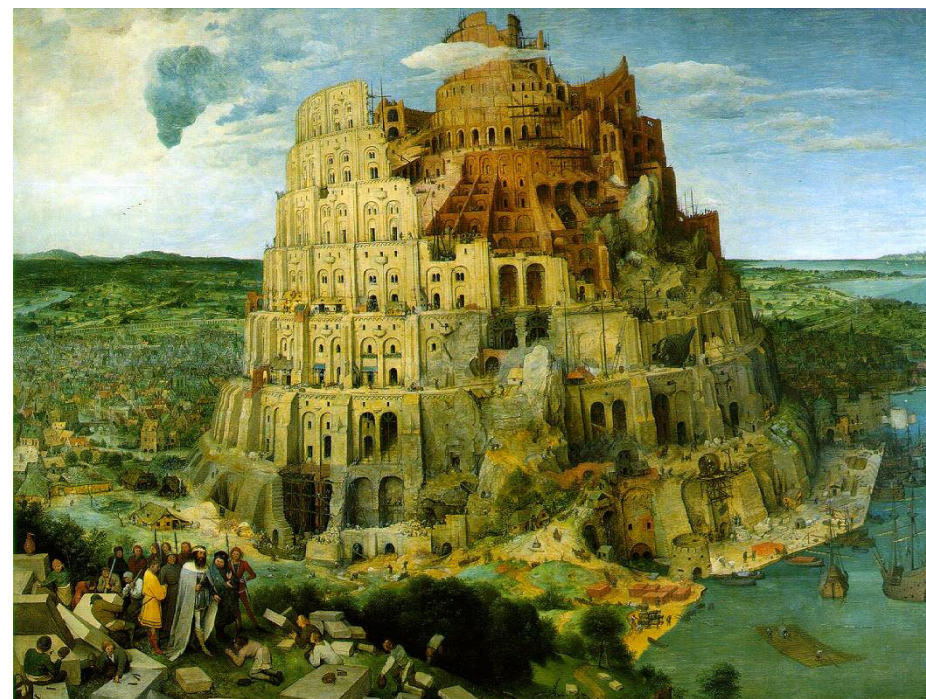


Automatic generation of three-dimensional atomic COoRdINates



http://www2.chemie.uni-erlangen.de/software/corina/free_struct.html

The screenshot shows the JME Molecule Editor interface. On the left is a vertical toolbar with buttons for elements: C, N, O, B, F, Cl, Br, I, P, X. The main window contains a 'SMILES string' field with the text [N]C@@H(C)C(=O)O. Below this is an 'Upload structure file' section with the text 'NOT YET SUPPORTED - COMING SOON'. There is a field for 'Please enter an identifier for your structure' containing 'L-alanine'. At the bottom, there are radio buttons for 'Choose a 3D structure viewer': 'automatically loaded (including PDB/SDF download option)' (selected) and 'external molecular viewer (e.g. RasMol, MDL Chime)'. Buttons for 'Transfer as SMILES', 'Clear Editor', 'Help!', and 'Generate 3D Structure' are visible.



BABEL A program designed to interconvert a number of file formats currently used in molecular modelling

Input type codes:

```
alc -- Alchemy file
prep -- AMBER PREP file
bs -- Ball and Stick file
bgf -- MSI BGF file
car -- Biosym .CAR file
boog -- Boogie file
cacprt -- Cacao Cartesian file
cadpac -- Cambridge CADPAC file
charmm -- CHARMM file
c3d1 -- Chem3D Cartesian 1 file
c3d2 -- Chem3D Cartesian 2 file
cssr -- CSD CSSR file
fdat -- CSD FDAT file
gstat -- CSD GSTAT file
dock -- Dock Database file
dpdb -- Dock PDB file
feat -- Feature file
fract -- Free Form Fractional file
gamout -- GAMESS Output file
gzmat -- Gaussian Z-Matrix file
gauout -- Gaussian 92 Output file
g94 -- Gaussian 94 Output file
gr96A -- GROMOS96 (A) file
gr96N -- GROMOS96 (nm) file
hin -- Hyperchem HIN file
sdf -- MDL Isis SDF file
m3d -- M3D file
macmol -- Mac Molecule file

macmod -- Macromodel file
micro -- Micro World file
mm2in -- MM2 Input file
mm2out -- MM2 Output file
mm3 -- MM3 file
mmads -- MMADS file
mdl -- MDL MOLfile file
molen -- MOLIN file
mopcrt -- Mopac Cartesian file
mopint -- Mopac Internal file
mopout -- Mopac Output file
pccmod -- PC Model file
pdb -- PDB file
psin -- PS-GVB Input file
psout -- PS-GVB Output file
msf -- Quanta MSF file
schakal -- Schakal file
shelx -- ShelX file
smiles -- SMILES file
spar -- Spartan file
semi -- Spartan Semi-Empirical file
spmm -- Spartan Mol. Mechanics file
mol -- Sybyl Mol file
mol2 -- Sybyl Mol2 file
wiz -- Conjure file
unxyz -- UniChem XYZ file
xyz -- XYZ file
xed -- XED file
```

BABEL A program designed to interconvert a number of file formats currently used in molecular modelling

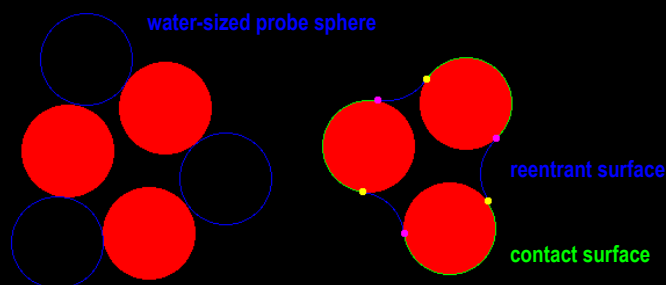
Output type codes:

```
diag -- DIAGNOTICS file
t -- Alchemy file
bs -- Ball and Stick file
bmin -- Batchmin Command file
cacprt -- Cacao Cartesian file
cacint -- Cacao Internal file
cache -- CAChe MolStruct file
c3d1 -- Chem3D Cartesian 1 file
c3d2 -- Chem3D Cartesian 2 file
d -- ChemDraw Conn. Table file
con -- Conjure file
contmp -- Conjure Template file
cssr -- CSD CSSR file
feat -- Feature file
fhz -- Fenske-Hall ZMatrix file
gamin -- Gamess Input file
gcart -- Gaussian Cartesian file
g -- Gaussian Z-matrix file
gotmp -- Gaussian Z-matrix tmplt file
hin -- Hyperchem HIN file
icon -- Icon 8 file

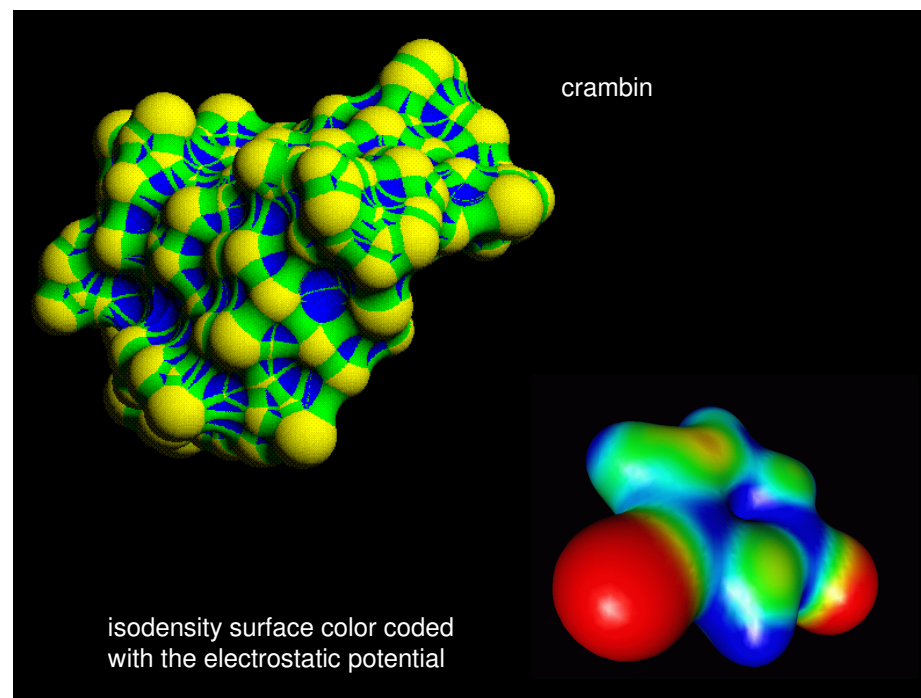
i -- IDATM file
macmol -- Mac Molecule file
k -- Macromodel file
micro -- Micro World file
mi -- MM2 Input file
mo -- MM2 Ouput file
mm3 -- MM3 file
mmads -- MMADS file
mdl -- MDL Molfile file
ac -- Mopac Cartesian file
ai -- Mopac Internal file
pc -- PC Model file
p -- PDB file
report -- Report file
spar -- Spartan file
mol -- Sybyl Mol file
mol2 -- Sybyl Mol2 file
maccs -- MDL Maccs file file
xed -- XED file
unxyz -- UniChem XYZ file
x -- XYZ file
```

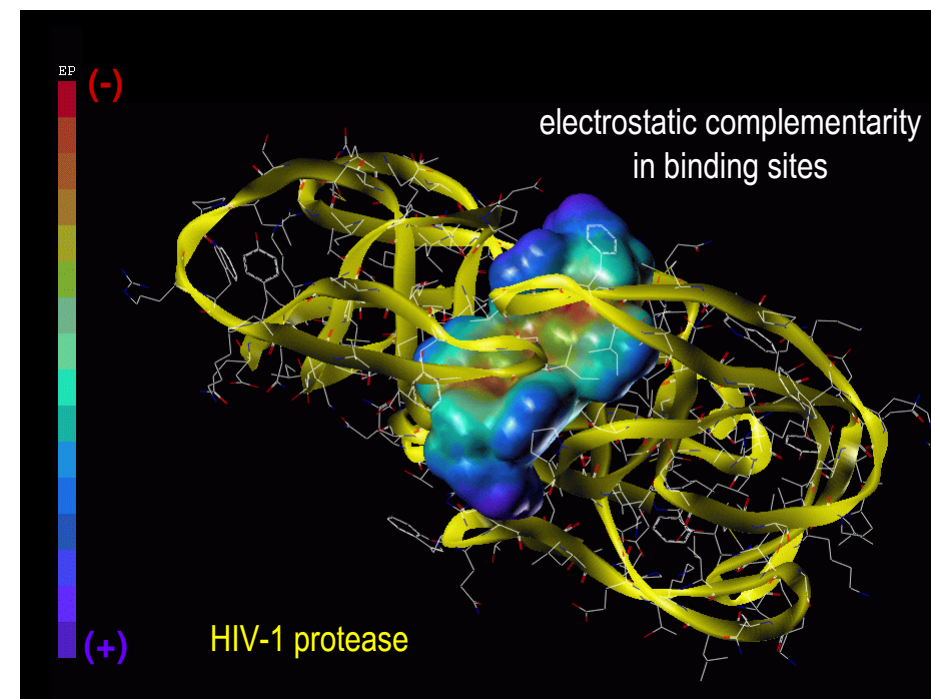
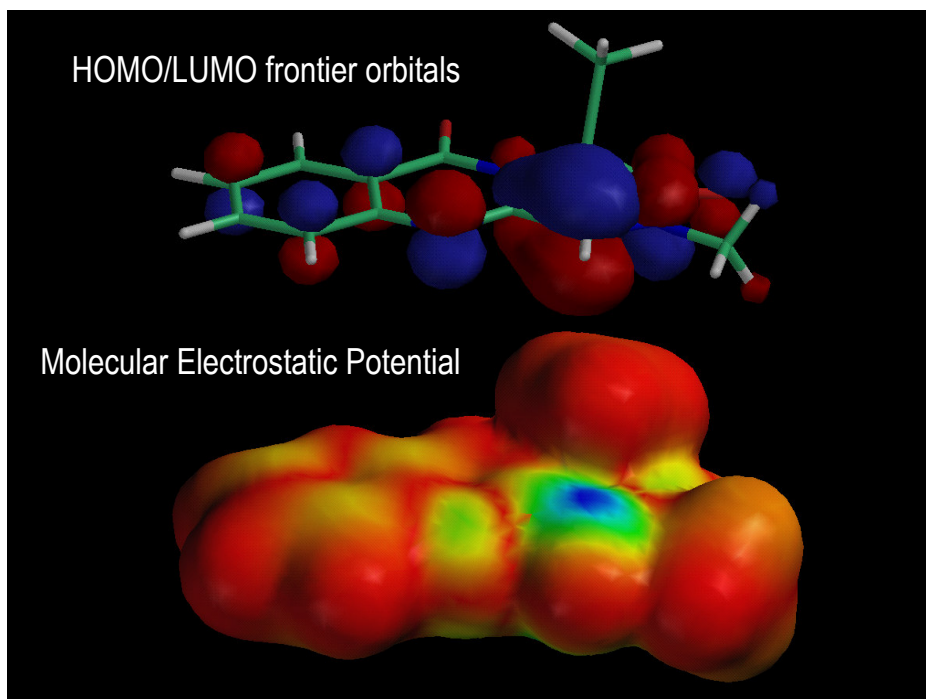
<ftp://ccl.osc.edu/pub/chemistry/software/UNIX/babel/>

Molecular Surfaces



The **solvent-accessible surface** (Lee and Richards, 1971) is traced out by the probe sphere center as it rolls over the molecule.





MOLECULAR MECHANICS (MM)

- ✓ A computational technique used to model the **conformational behaviour** and **energetic properties** of molecules.
- ✓ The molecule is treated at the **atomic level**, i.e. the electrons are not treated explicitly.
- ✓ MM uses an **Energy Function**, defined so that given a particular conformation, (i.e. given a set of spatial coordinates for all the atoms) the energy of the molecule can be calculated.
- ✓ The energy function is **empirical**, i.e. it is not entirely derived from rigorous theories.
- ✓ The energy function makes a distinction between '**bonded**' and '**non-bonded**' interactions.

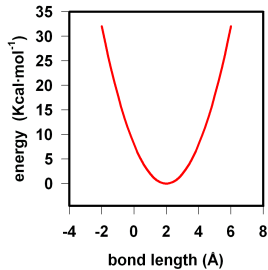
Mecánica Molecular

$$E_{potencial} = E_{enlazada} + E_{no-enlazada}$$

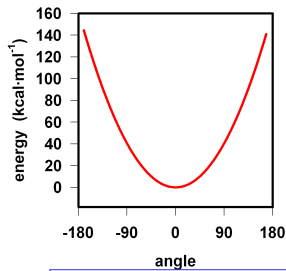
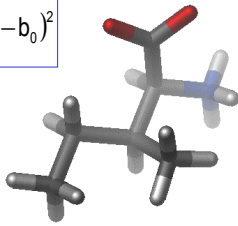
$$E_{enlazada} = \sum_i E_{enlaces} + \sum_i E_{ángulos} + \sum_i E_{diedros}$$

$$E_{no-enlazada} = \sum_i E_{electrostática} + \sum_i E_{van\ der\ Waals}$$

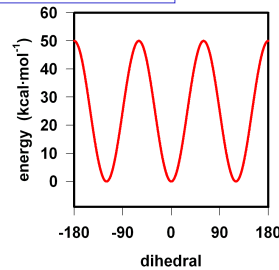
TÉRMINOS ENLAZADOS



$$E_{\text{enlaces}} = \sum_{\text{enlaces}} \frac{1}{2} k_b (b - b_0)^2$$



$$E_{\text{ángulos}} = \sum_{\text{ángulos}} \frac{1}{2} k_\theta (\theta - \theta_0)^2$$

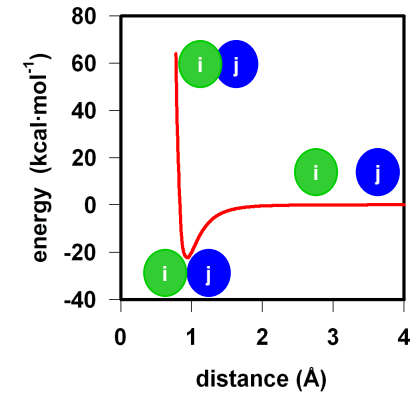
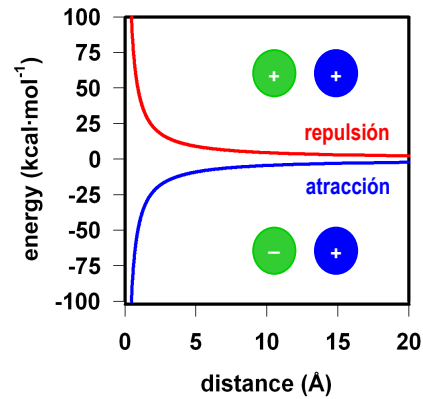


$$E_{\text{dihedros}} = \sum_{\text{dihedros}} \frac{1}{2} k_\phi [1 + \cos(\phi - \phi_0)]$$

TÉRMINOS NO-ENLAZADOS

$$E_{\text{electrostática}} = \frac{1}{4\pi\epsilon_0\epsilon} \sum_{ij} \frac{q_i q_j}{r_{ij}}$$

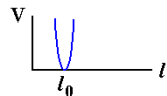
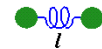
$$E_{\text{Lennard-Jones}} = \sum_{ij} \frac{A_{ij}}{r_{ij}^{12}} - \frac{B_{ij}}{r_{ij}^6}$$



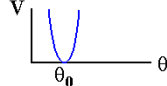
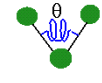
Empirical Potential Energy Function

Resumen de interacciones incluidas en un **campo de fuerzas** representativo de mecánica molecular

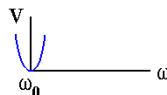
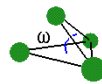
Bonds



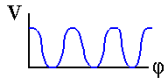
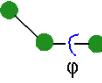
Angles



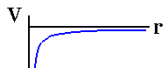
Improper
Dihedrals



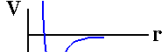
Torsions



Electrostatics



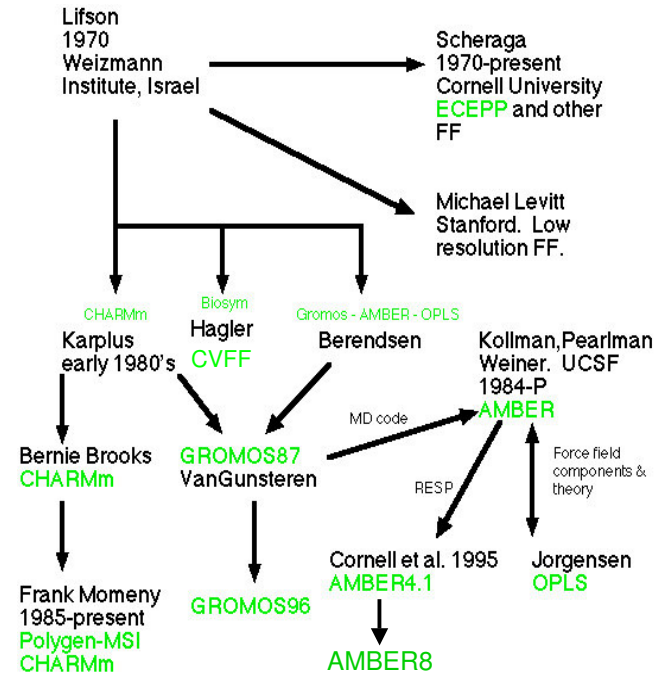
van der Waals



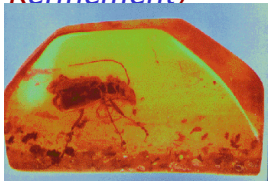
$$V = \sum_{\text{enlaces}} k_b (b - b_0)^2 + \sum_{\text{ángulos}} k_\theta (\theta - \theta_0)^2 + \sum_{\text{dihedros } n=1}^N K_\phi^{(n)} [1 + \cos(n\phi - \delta)] + \sum_{\text{impropios}} K_\omega (\omega - \omega_0)^2 + \sum_{i,j} 4\epsilon_{ij} \left[\left(\frac{\sigma_{ij}}{r_{ij}} \right)^{12} - \left(\frac{\sigma_{ij}}{r_{ij}} \right)^6 \right] + \sum_{i,j} \left(\frac{q_i q_j}{D r_{ij}} \right)$$

✓ La función empírica de energía potencial es **diferenciable** con respecto a las coordenadas atómicas.

✓ Esto proporciona el valor y la dirección de la **fuerza** que actúa sobre cada átomo y puede así utilizarse en una **simulación de dinámica molecular**.



AMBER (Assisted Model Building with Energy Refinement)



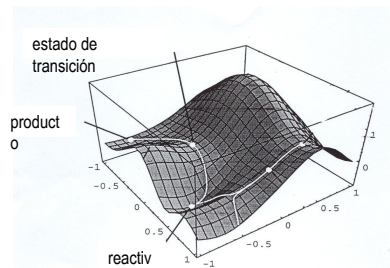
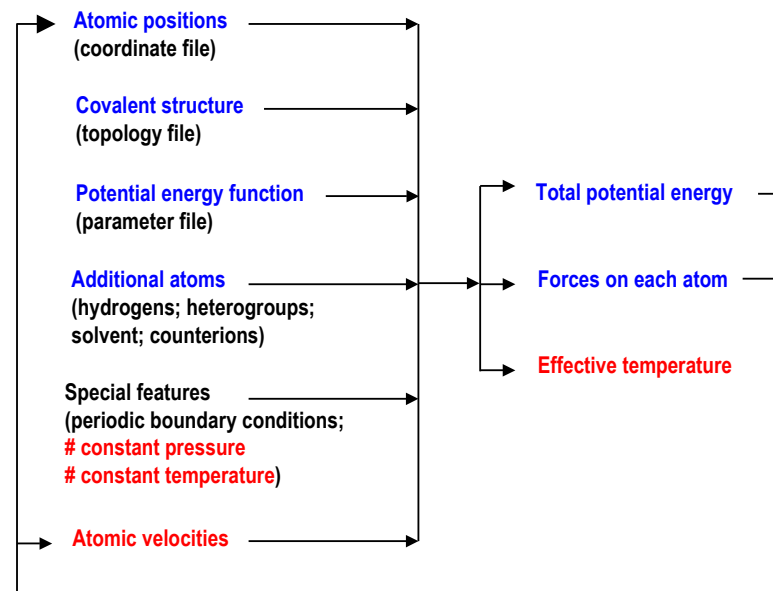
CHARMm® (Chemistry at HARvard Macromolecular Mechanics)

CVFF (Consistent-Valence Force Field)

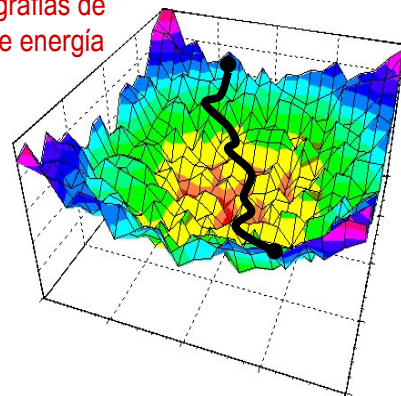
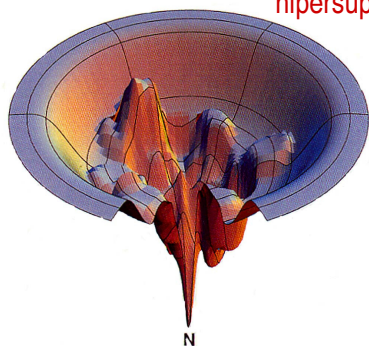
GROMOS (GRONingen MOlecular Simulation package)

OPLS (Optimized Potentials for Liquid Simulations)

ALGORITHMS FOR ENERGY MINIMIZATION AND MOLECULAR DYNAMICS



Ejemplos de topografías de hipersuperficies de energía



Dinámica Molecular

Segunda ley de Newton:

$$-\frac{dE}{dx} = F = m \cdot a = m \cdot \frac{v}{t} = m \cdot \frac{d^2x}{dt^2}$$

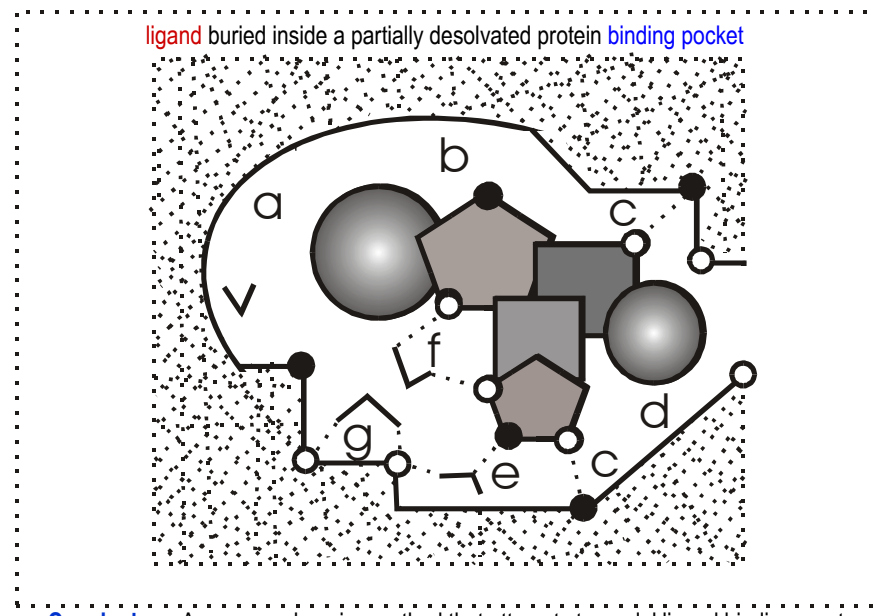
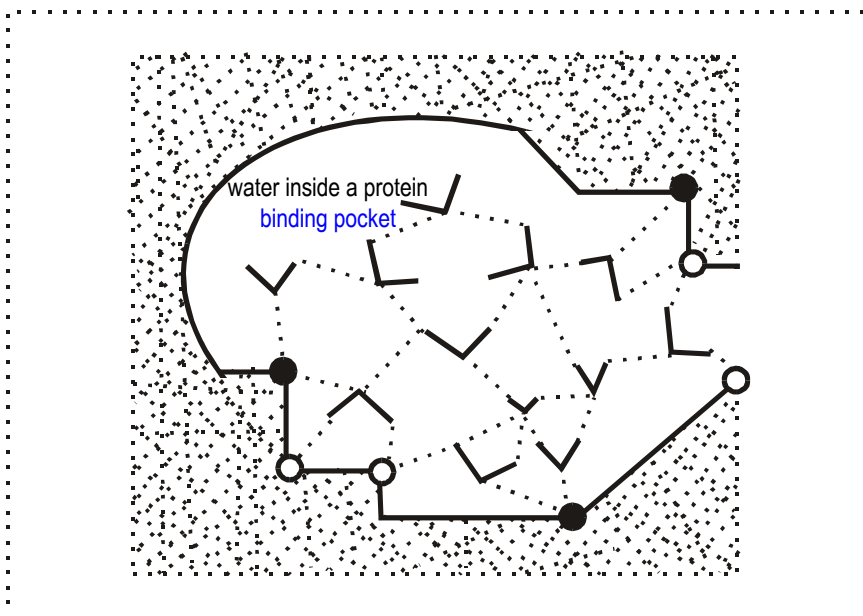
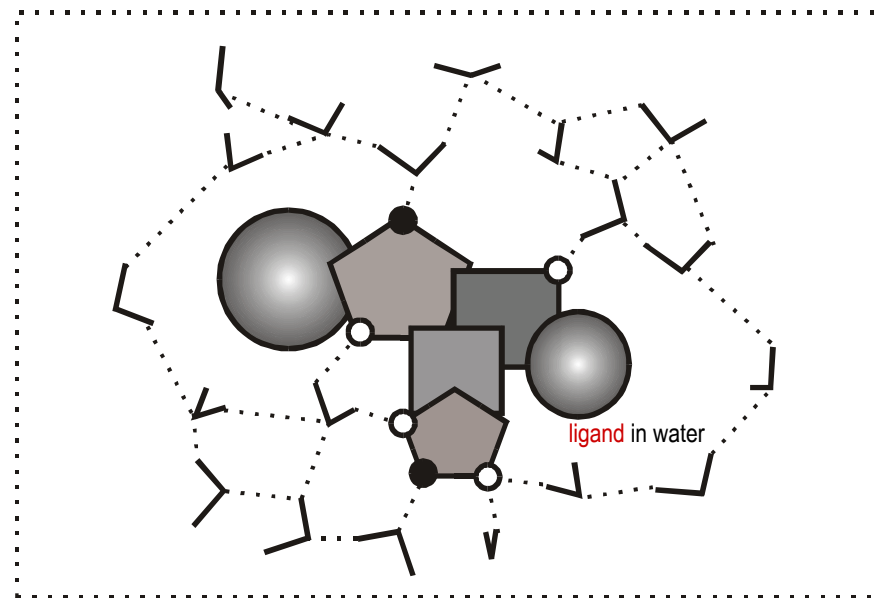


Las simulaciones de Dinámica Molecular permiten el estudio de procesos dinámicos complejos que ocurren en los sistemas biológicos, por ejemplo:

- ✓ Estabilidad de proteínas
- ✓ Cambios conformacionales
- ✓ Plegamiento de proteínas
- ✓ Reconocimiento molecular: ligandos, proteínas, ADN...
- ✓ Transporte de iones en sistemas biológicos

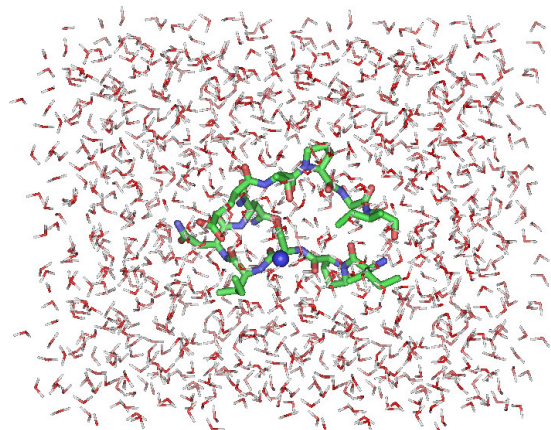
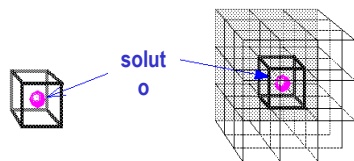
y proporcionan un medio valioso para llevar a cabo estudios de:

- determinación de estructuras por difracción de rayos X y espectroscopía de RMN
- diseño de nuevos fármacos

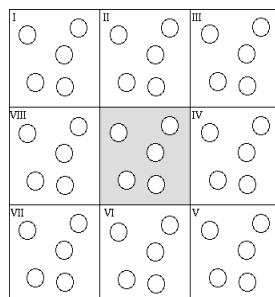


Conclusion: Any comprehensive method that attempts to model ligand binding must also consider the energy of solvation and entropic contributions to the binding process.

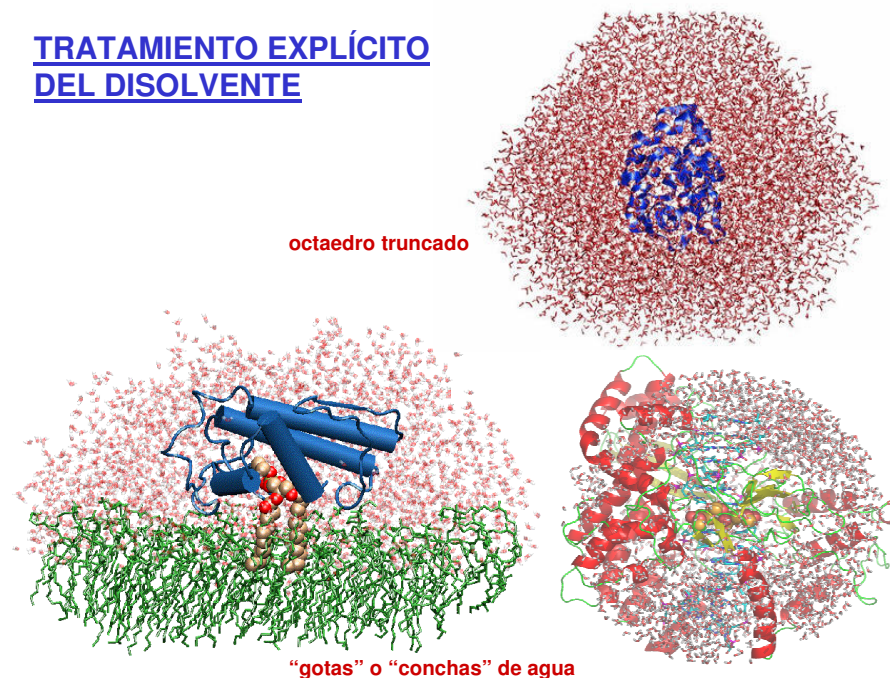
TRATAMIENTO EXPLÍCITO DEL DISOLVENTE



CONDICIONES DE LÍMITE PERIÓDICO

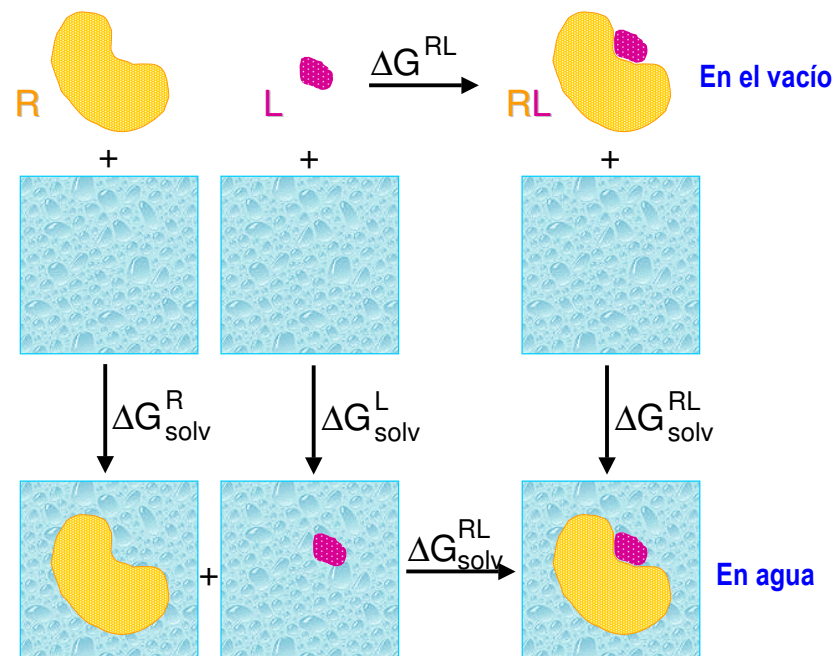


TRATAMIENTO EXPLÍCITO DEL DISOLVENTE



octaedro truncado

"gotas" o "conchas" de agua



"DelPhi - A Macromolecular Electrostatics Modelling Package":

Kim A. Sharp, Anthony Nicholls & Barry Honig

Department of Biochemistry and Molecular Biophysics, Columbia University, New York

- Klapper, I.; Hagstrom, R.; Fine, R.; Sharp, K.; Honig, B. "Focusing of Electric Fields in the Active Site of Cu-Zn Superoxide Dismutase: Effects of Ionic Strength and Amino-acid Modification." *Proteins* (1986) 1, 47-59.

- Gilson, M. K.; Sharp, K. A.; Honig, B. H. "Calculating the Electrostatic Potential of Molecules in Solution: Method and Error Assessment" *J. Comput. Chem.* (1987) 9, 327-335.

- Gilson, M. K.; Honig, B. "Calculation of the Total Electrostatic Energy of a Macromolecular System: Solvation Energies, Binding Energies, and Conformational Analysis." *Proteins* (1988) 4, 7-18.

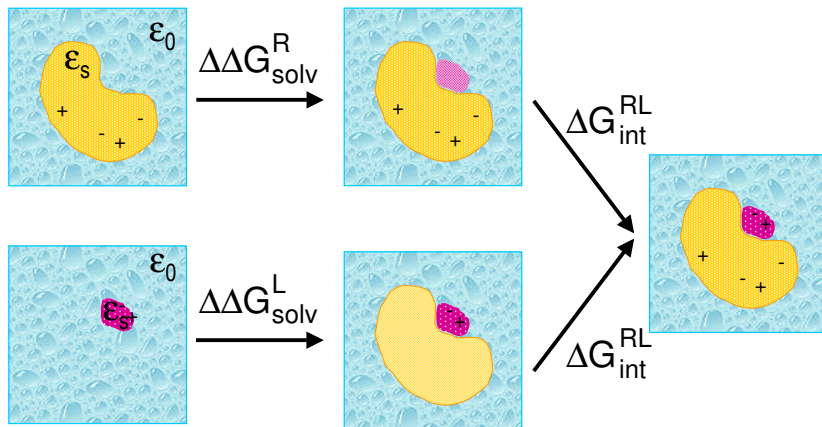
- K. Sharp, K.; Honig, B. "Electrostatic Interactions in Macromolecules: Theory and Applications." *Ann. Rev. Biophys. Biophys. Chem.* (1990) 19, 301-332.

- Nicholls, A.; Honig, B. "A Rapid Finite Difference Algorithm, Utilizing Successive Over-Relaxation to Solve the Poisson-Boltzmann Equation." *J. Comput. Chem.* (1991) 12, 435-445.

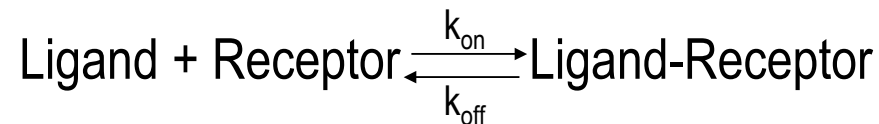
The original reference to the use of the finite difference method for macromolecular electrostatics is: J. Warwicker and H. C. Watson, *J. Mol. Biol.* (1982) 157, 671.

Ecuación de Poisson: $\nabla^2 \phi(r) = -\frac{4\pi\rho(r)}{\epsilon}$

Ecuación de Poisson-Boltzmann: $\nabla \cdot [\epsilon(r)\nabla\phi(r)] - k'\phi(r) = -4\pi\rho(r)$



Affinity vs. Specificity



$$K_d = \frac{k_{\text{off}}}{k_{\text{on}}} = \frac{[\text{Ligand}] [\text{Receptor}]}{[\text{Ligand-Receptor}]}$$

$$\Delta G = \Delta H - T\Delta S$$

Binding constant

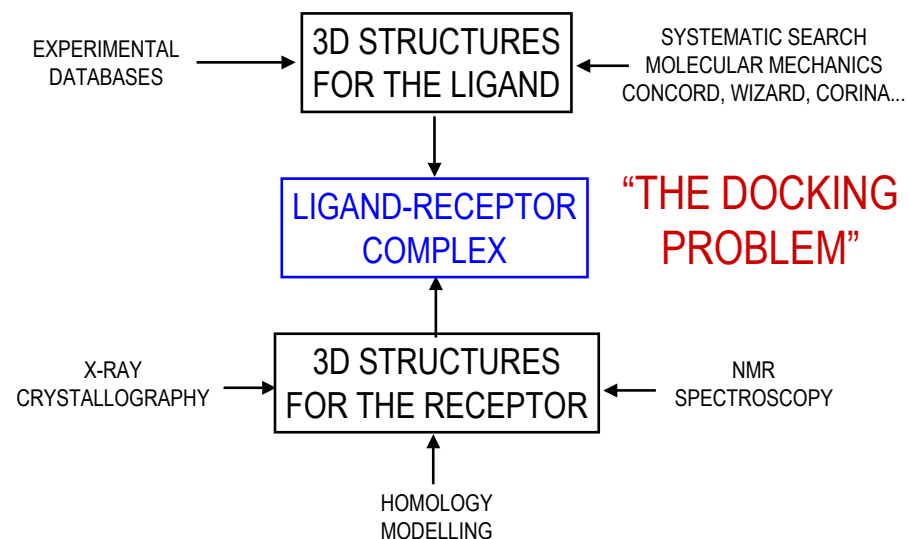
Binding energy

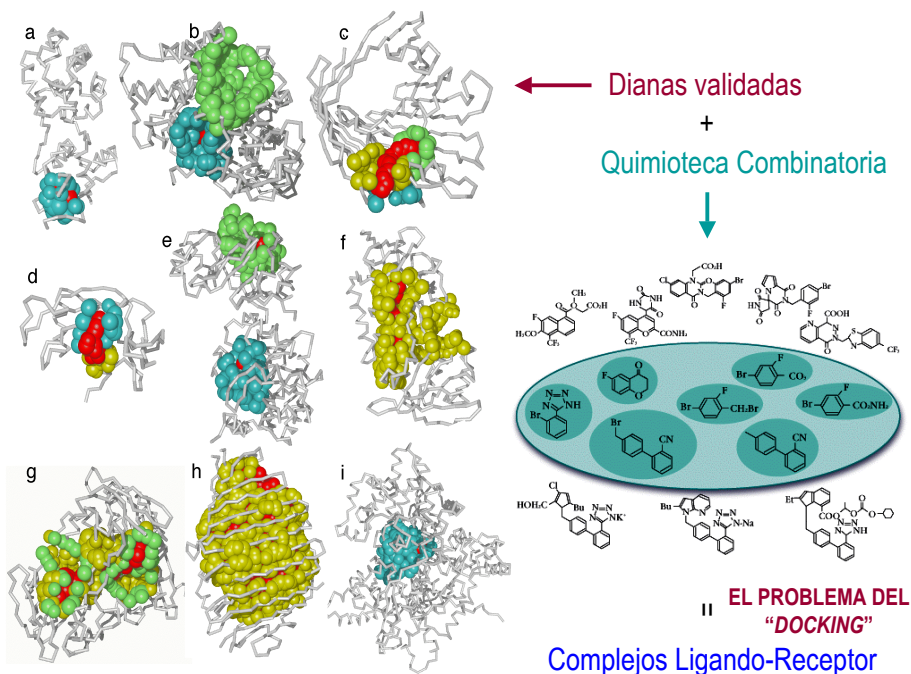
$$\Delta K_d$$

$$\Delta G \text{ (kcal/mol)}$$

2x	0.5
5x	1.0
13x	1.5
29x	2.0
68x	2.5
158x	3.0

$$\Delta G = 2.303 RT \log K_d$$





CASTp A Server for Identification of Protein Pockets & Cavities

- Identifies all pockets and cavities.
- Measures the volume and area analytically.

CASTp

1dff

Pocket Information

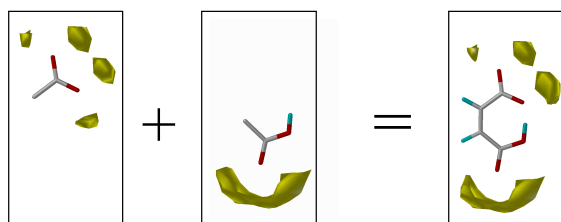
ID	AREA	VOL
21	346.0	604.3
20	151.3	231.1
19	114.3	118.3
18	101.4	74.1
17	55.4	73.4
16	57.4	78.3
15	45.1	38.3
14	35.4	34.0
13	48.0	23.3
12	38.0	25.5
11	41.8	21.8
10	20.5	14.8
9	35.2	18.6
8	14.5	8.5
7	28.9	22.7
6	34.1	18.6
5	32.5	19.8
4	15.8	9.8
3	28.8	12.9

Pocket Detail

Num	Atom	AA	Chain
77	CD1	LEU	
77	CD2	LEU	
80	CE	LYS	
80	NZ	LYS	
101	O	PRO	

http://cast.engr.uic.edu/cast/

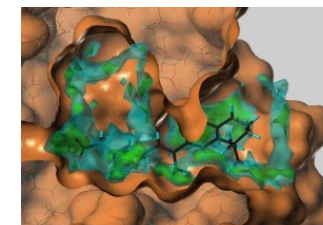
SuperStar



- Calculate binding positions for specific probe atoms in protein active sites
- Identify functional groups in binding-site
- Look up relevant IsoStar scatterplots and overlay on functional groups
- Contour - combining by taking products

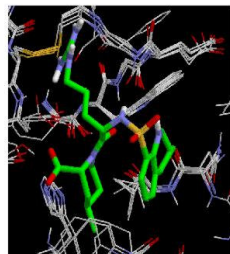
SuperStar Features

map for aromatic CH carbon probe generated at the binding site of the protein-ligand complex 1CPS.



- Cavity detection
- Surface or pharmacophore point display
- Metal coordination
- Hyperlinking to IsoStar scatterplots
- Choice of CSD- or PDB-based maps
- Gaussian fits

Relibase



Relibase: A program for searching protein-ligand databases.

Version 4.0, October 2000

Relibase is copyright Manfred Hendlich 1994-1999 and Cambridge Crystallographic Data Centre 1999, 2000.

[Install 3D visualization software](#)

Comments/bugs/queries to:

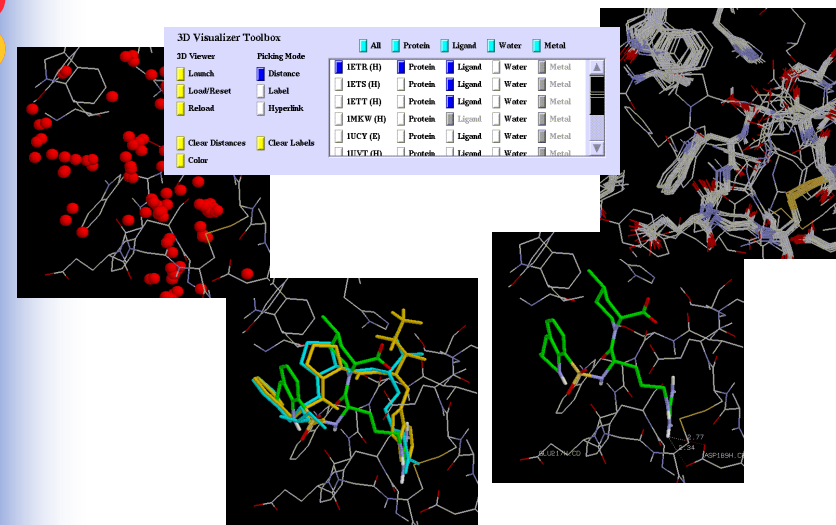
Cambridge Crystallographic Data Centre
12 Union Road
Cambridge CB2 1EZ
United Kingdom
tel: (44) (1223) 336022
e-mail: support@ccdc.cam.ac.uk

<http://relibase.ccdc.cam.ac.uk/>

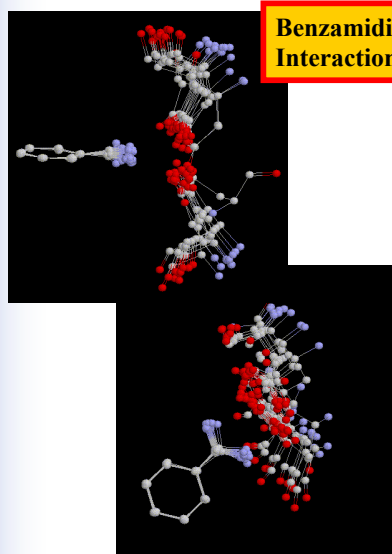
<http://relibase.ebi.ac.uk/>

<http://relibase.rutgers.edu/>

Binding Site Superposition

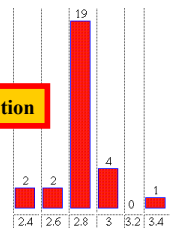


Analysis of 3D Queries

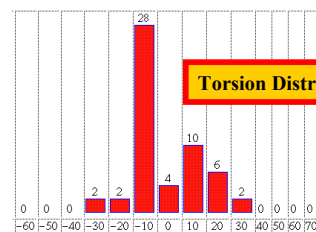


Benzamidine-Carboxylate Interactions

Distance Distribution



Torsion Distribution



Relibase+

- Protein-ligand database system
- Based on original software developed by Manfred Hendlich and colleagues at Merck and Marburg University
- Enables searching of PDB and of in-house proprietary databases

Relibase: <http://relibase.ccdc.cam.ac.uk/>

Some Relibase+ Options

- Text searching
- Sequence searching
- 2D substructure and similarity searching
- 3D substructure searching
- Logical combination of hit lists
- Searching for intermolecular interactions
- Auto-superposition of similar binding sites
- Scripting facility based on Python

LIGPLOT

<http://www.biochem.ucl.ac.uk/bsm/ligplot/ligplot.html>

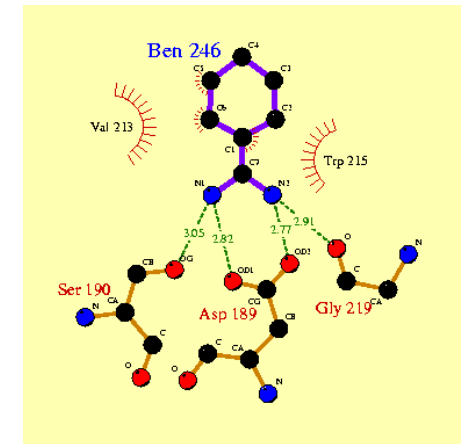
Program for automatically plotting protein-ligand interactions (by A. Wallace & R. Laskowski)

Automatically generates schematic diagrams of protein-ligand interactions for a given PDB file.

hydrogen bonds: dashed lines between the atoms involved.

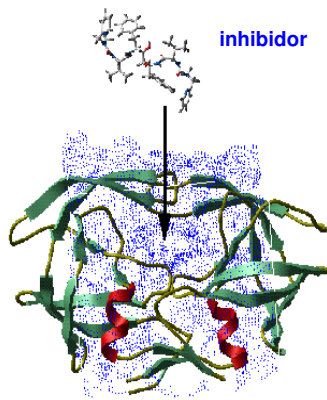
hydrophobic contact: an arc with spokes radiating towards the ligand atoms they contact. The contacted atoms are shown with spokes radiating back.

Atom accessibilities can also be depicted; the ligand atoms can be colour-coded to indicate their accessibility to solvent.

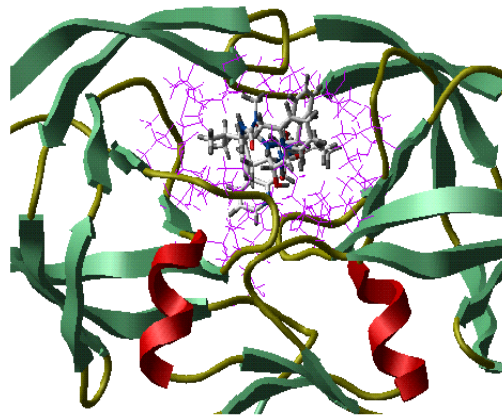


benzamidine (PDB code 2TBS)

ACOPLAMIENTO LIGANDO-RECEPTOR (DOCKING



Proteasa de VIH-1



Complejo enzima-inhibidor

Virtual (“in silico”) screening

Docking/scoring programs

Docking engines: search the conformational space in the binding site

Scoring functions: discrimination of correctly docked from misdocked conformations

MOLECULAR DOCKING

□ SYSTEMATIC SEARCH (*brute force algorithm*):

All binding orientations of all conformers of the ligand and the receptor (impractical for most situations).

□ AUTOMATED SEARCH:

GEOMETRIC METHODS: Matching of ligand and receptor site descriptors (descriptors, grids, fragments...).

FORCE FIELD METHODS: Minimizing the ligand-receptor interaction energy - Molecular dynamics and Monte Carlo simulations.

“GRID: A Computational Procedure for Determining Energetically Favorable Binding Sites on Biologically Important Macromolecules”

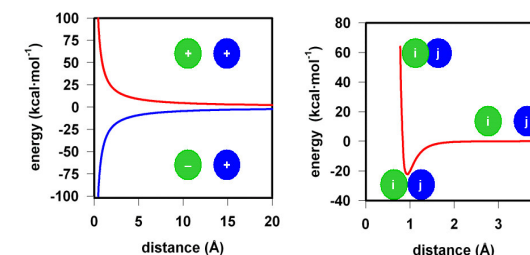
Peter Goodford, Oxford University

J. Med. Chem. 28, 849-857 (1985)

ibid. 32, 1083-1094 (1989); 36, 140-147 (1993); 36, 148-156 (1993)



<http://www.moldiscovery.com/>



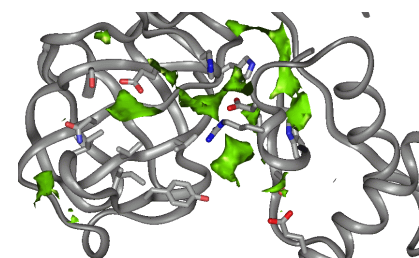
Probe selection...

symbol	description	selected
1 OH2	Water	
2 DRY	The Hydrophobic Probe	
3 H	Hydrogen	
4 C3	Methyl CH3 group	
5 C1=	sp2 CH aromatic or vinyl	
6 N#	sp N with lone pair	
7 N=	sp2 N with lone pair	
8 N:	sp3 N with lone pair	
9 N-	Anionic tetrazole N	
10 N1	Neutral flat NH eg amide	
11 N1+	sp3 amine NH cation	

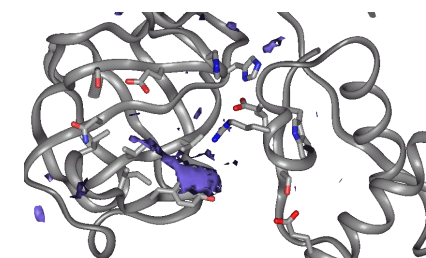
OK Cancel

Structure 1 (glucose) Field: 1

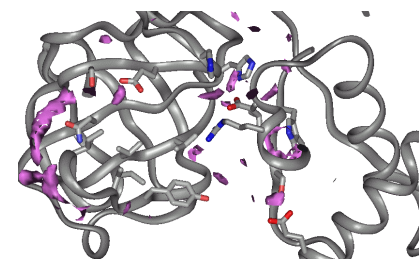
File View Data Molecules



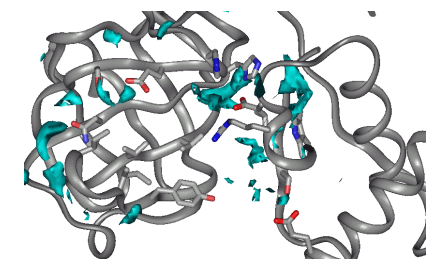
Aromatic carbon probe
Grid point value range: -5.45 to 5.0 kcal/mol
Contour level: -2.5 kcal/mol



Hydrophobic probe
Grid point value range: -2.86 to 0.0 kcal/mol
Contour level: -1.0 kcal/mol



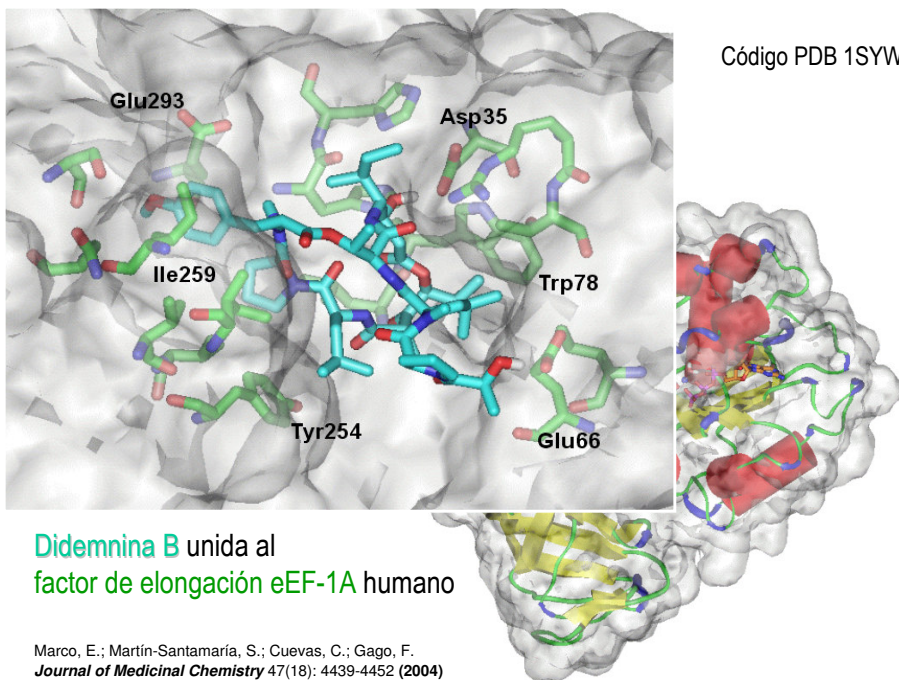
Carbonyl oxygen probe
Grid point value range: -8.03 to 5.0 kcal/mol
Contour level: -5.0 kcal/mol



Hydroxyl oxygen probe
Grid point value range: -12.30 to 5.0 kcal/mol
Contour level: -7.0 kcal/mol



<http://www.moldiscovery.com/>



Why Use Molecular Docking?

- Most detailed representation of binding site
 - overcomes simplifications of pharmacophores
 - identifies both conservative and novel solutions
 - provides impetus for *de novo* design/optimisation
- Broad range of analyses applicable
 - diverse scoring/selection criteria
- Quality/throughput of available methods
 - good enough, despite technical limitations

“THE DOCKING PROBLEM”

- SITE/LIGAND REPRESENTATION (treatment of H atoms?)
- JUXTAPOSITION OF THE LIGAND AND SITE FRAMES OF REFERENCE (docking engine)
- EVALUATION OF COMPLEMENTARITY (scoring functions)

AIM: To obtain the lowest free energy structure(s) for the receptor-ligand complex

Examples of docking algorithms

Rigid ligand:

Fast shape matching (DOCK)

Flexible ligand:

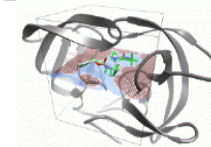
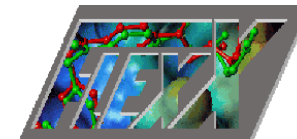
Fast shape matching (DOCK 4.0)

Incremental construction (FlexX)

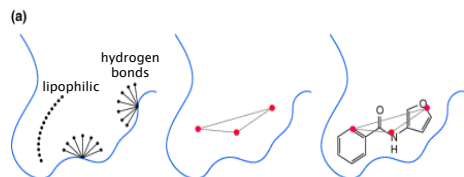
Simulated annealing (AutoDock 2.4)

Monte Carlo simulations (MCDOCK)

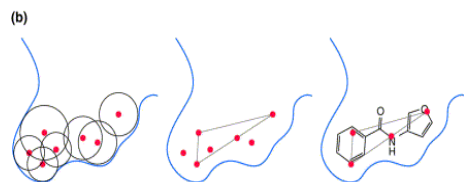
Genetic algorithm (AutoDock 3.0, GOLD, GAMBLER)



FlexX
algorithm



DOCK
algorithm



surface of the receptor pocket

FlexX matches triangles of interaction sites onto complementary ligand atoms.

DOCK fills the binding site with spheres, and sphere centers are then matched to the ligand atoms to determine plausible ligand-receptor complexes.

PROGRAM DOCK

"A Geometric Approach to Macromolecule-Ligand Interactions"

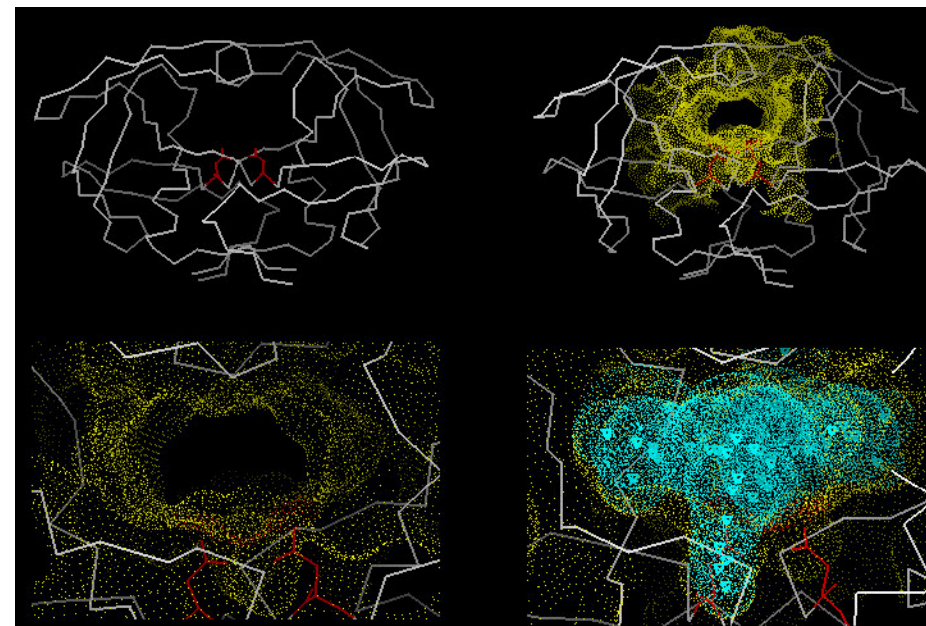
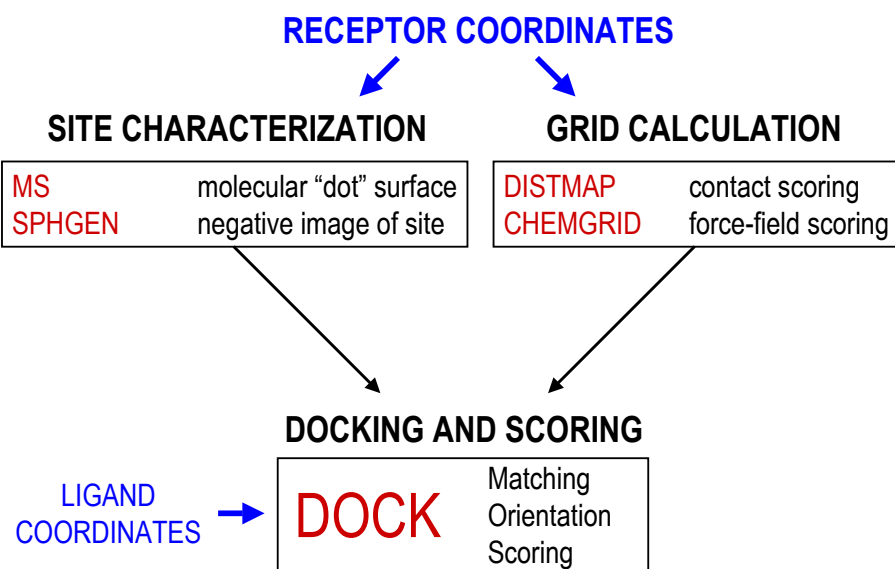
I. D. Kuntz, J. M. Blaney, S. J. Oatley, R. Langridge, T. E. Ferrin
J. Mol. Biol. 161, 269-288 (1982)

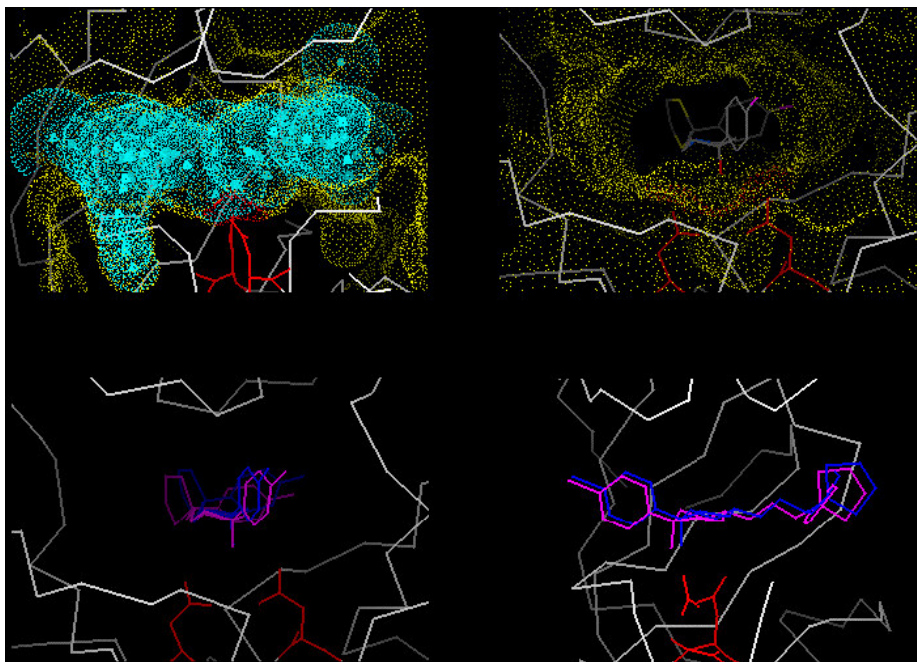
"Using Shape Complementarity as an Initial Screen in Designing Ligands for a Receptor Binding Site of Known Three-Dimensional Structure"

R. L. DesJarlais, R. P. Sheridan, G. L. Seibel, J. S. Dixon, I. D. Kuntz, R. Venkataraghavan
J. Med. Chem. 31, 722-729 (1988)

"Automated Docking with Grid-Based Energy Evaluation"

E. C. Meng, B. K. Soichet, I. D. Kuntz
J. Comp. Chem. 13, 505-524 (1991)





Docking accuracy
 [Rms deviations (non hydrogen atoms, in Å) from the X-ray pose]
 (top solution of each docking tool)

ligand	Docking method		
	DOCK	FlexX	GOLD
deoxythymidine	0.82	0.78	0.72
5-iododeoxyuridine	9.33	1.03	0.77
5-iodouracil-anhydrohexitol	1.16	0.88	0.63
dhbt (not publicly available)	2.02	3.65	0.93
6-(3-hydroxy-propyl-thymine)	1.02	4.18	0.49
6-[6-hydroxymethy-5-methyl-2,4-dioxo-hexahydro-pyrimidin-5-yl-methyl]-5-methyl-1H-pyrimidin-2,4-dione	9.62	13.30	2.33
(North)-methanocarbothymidine	7.56	1.11	1.19
aciclovir	3.08	2.71	2.74
ganciclovir	3.01	6.07	3.11
penciclovir	4.10	5.96	3.01

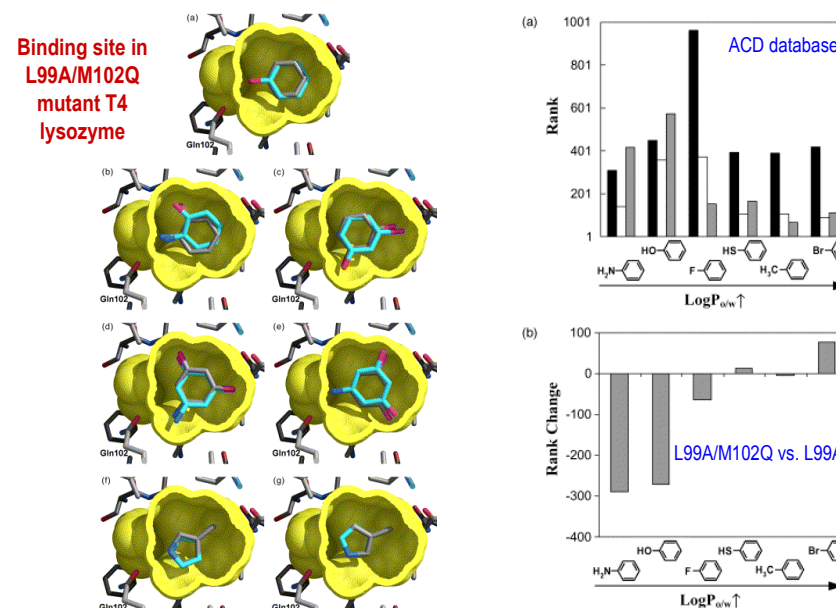
Only one set of protein (TK) coordinates used: pdb code 1kim

Scoring functions

Knowledge-based: statistical analysis of 3D complex structures to derive a sum of *potentials of mean force* between receptor and ligand atoms

Force field-based: calculation of van der Waals and electrostatic interaction energies between the receptor and the ligand atoms

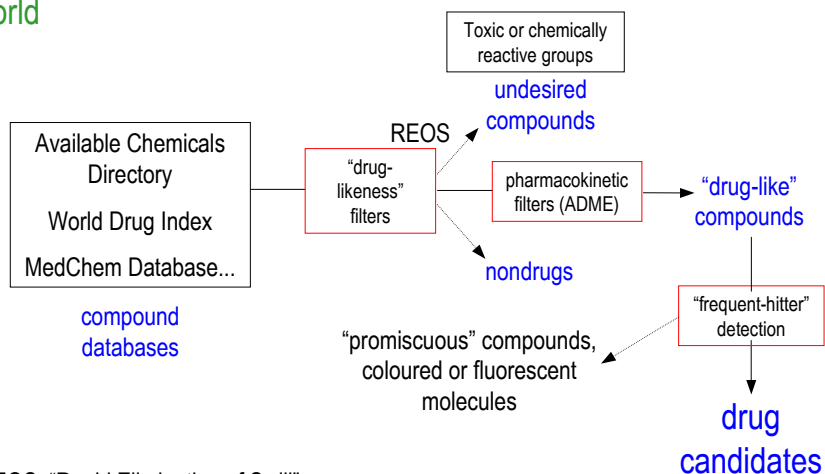
Empirical: the binding free energy is broken down into a number of different *weighted* contributions (supposed to be additive: number of hydrogen bonds, ionic interactions, apolar contacts, entropy penalties...)



Wei BQ, Baase WA, Weaver LH, Matthews BW, Shoichet BK.
 A model binding site for testing scoring functions in molecular docking.
J. Mol. Biol. (2002) 322:339-355

In silico VIRTUAL SCREENING and FOCUSED LIBRARY DESIGN

Near-perfect structures in an imperfect world

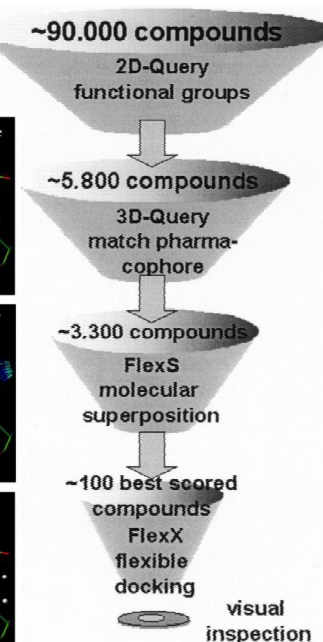
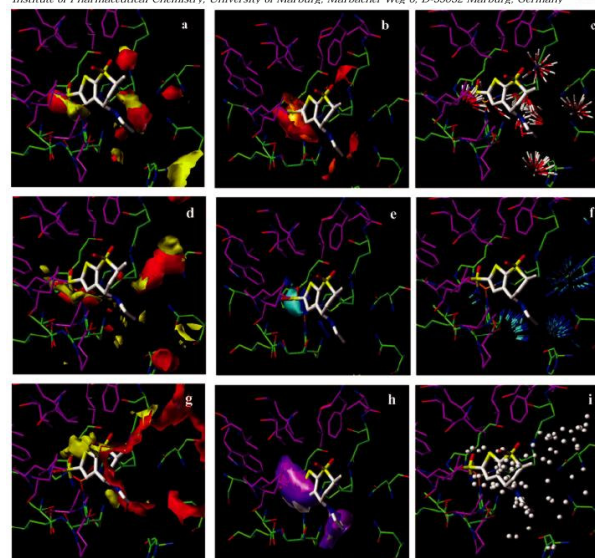


REOS: "Rapid Elimination of Swill"

Successful Virtual Screening for Novel Inhibitors of Human Carbonic Anhydrase: Strategy and Experimental Confirmation

Sven Grüneberg,¹ Milton T. Stubbs, and Gerhard Klebe^{*}

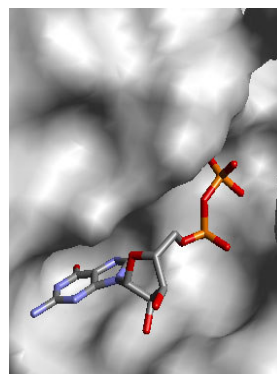
¹Institute of Pharmaceutical Chemistry, University of Marburg, Marbacher Weg 6, D-35032 Marburg, Germany



University of Oxford
Screensaver Lifesaver
searching for anti-cancer drugs by distributed computational chemistry



<http://www.chem.ox.ac.uk/ccdd/ccdd.html>



Superoxide dismutase	Vascular Endothelial Growth Factor
RAS proteins	Insulin Tyrosine Kinase
Cyclooxygenase (COX-2)	c-ABL Tyrosine Kinase
Fibroblast Growth Factor Receptor	CDK-2
RAF	Farnesyltransferase
Protein-Tyrosine-Phosphatase 1B	VEGFr1

fightAIDS@home the Olson laboratory
computing toward a cure
powered by ENTROPiA™

<http://FightAidsathome.scripps.edu>

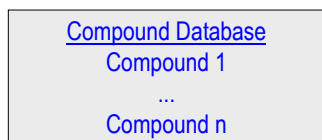


Dr. Garrett Morris
fightAIDS@home computing toward a better inhibitor powered by AUTODOCK

Two molecular docking visualizations showing ligands bound to protein active sites. The top image shows a ligand in a protein pocket with Energy = -36.84 kcal/mol, RMSD = 0.43 Angstrom, and C-offinity = 5.0. The bottom image shows a different ligand with Energy = -76.69 kcal/mol, RMSD = 0.86 Angstrom, and C-offinity = 5.0.

Applications of Ligand-Protein Docking in Drug Design

Existing methods
Given a protein, find potential binding ligands from a chemical database

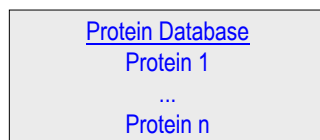


Protein

Successfully docked compounds as putative ligands

Science (1992) 257: 1078

New method
Given a ligand, find potential protein targets from a protein database

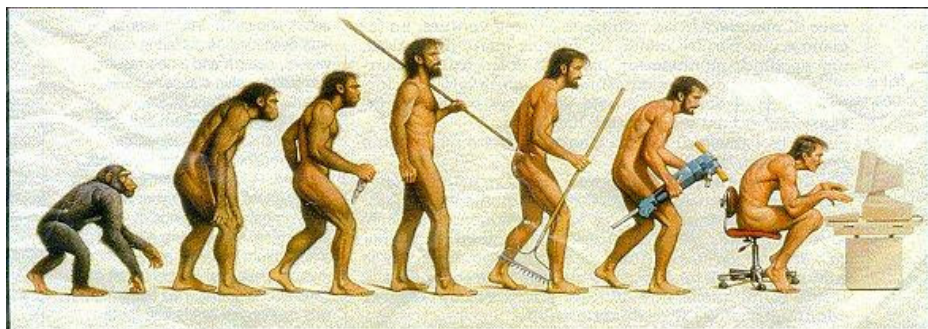
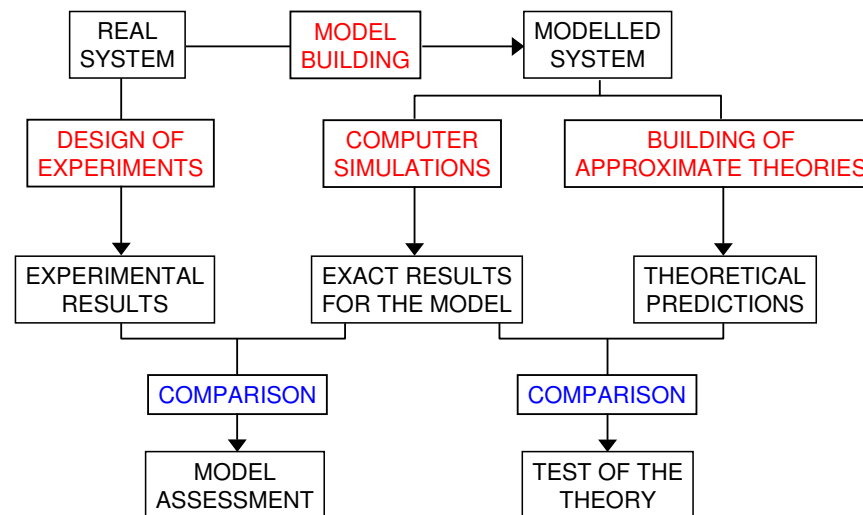


Ligand

Successfully docked proteins as putative targets

Proteins (2001) 43: 217

CONNECTION BETWEEN EXPERIMENT, THEORY AND COMPUTER SIMULATION



Somewhere, something went terribly wrong

QUESTIONS WELCOME

PREGUNTAS, POR FAVOR

E-mail: federico.gago@uah.es